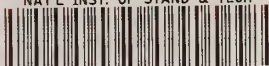


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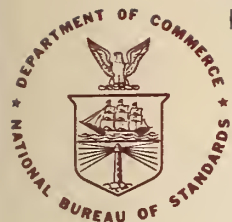












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## Four Mathematical Models for the Prediction of LNG Densities

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Robert D. McCarty

Thermophysical Properties Division  
National Engineering Laboratory  
National Bureau of Standards  
Boulder, Colorado 80303



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# FOUR MATHEMATICAL MODELS FOR THE PREDICTION OF LNG DENSITIES\*

Robert D. McCarty

Thermophysical Properties Division  
National Engineering Laboratory  
National Bureau of Standards  
Boulder, Colorado 80303

Four mathematical models of the equation of state for LNG like mixtures are presented. The four models include an extended corresponding states model, a cell model, a hard sphere model and a revised Klosek and McKinley model. Each of the models has been optimized to the same experimental data set which included data for pure nitrogen, methane, ethane, propane, iso and normal butane, iso and normal pentane and mixtures thereof. For LNG like mixtures (mixtures of the orthobaric liquid state at temperatures of 120 K or less and containing at least 60% methane, less than 4% nitrogen, less than 4% each of iso and normal butane and less than 2% total of iso and normal pentane), all of the models are estimated to predict densities to within 0.1% of the true value. The revised Klosek and McKinley model is valid only for mixtures within the range of temperature and composition specified above while the other three models are valid for a broader range of pressure, temperature and composition. The experimental PVTx data set used in the optimization together with comparisons are given and listings of computer programs for each of the models are included.

Key words: Cell model; comparisons; computer programs; corresponding states; equation of state; hard sphere; LNG; mixtures; PVTx data; revised Klosek and McKinley.

## 1. INTRODUCTION

The purpose of this report is to present in final form the four mathematical models which were optimized to the experimentally determined orthobaric liquid

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PVTx data of Miller and Hiza [25], Haynes and Hiza [12], Haynes, et al. [11], Hiza, et al. [14], Hiza and Haynes [15] and Haynes [9]. Interim results of the project have been reported by Haynes, et al. [13] and by McCarty [23]. The models reported in these two interim publications differ only slightly from those presented here. A companion archival document, McCarty [24] with the same results but in much less detail has been submitted to the Journal of Chemical Thermodynamics. The intent of the documentation here is more in the vein of a user's handbook. The above experimental data are for the liquid phase of nitrogen, methane, ethane, propane, normal and isobutane and various mixtures thereof.

The goal of the project (the project included the above referenced experimental work) was to produce one or more computer models which would predict the density of LNG to within 0.1% of the true value from a knowledge of the temperature, pressure and composition of the LNG. At the beginning of this study LNG was defined as mixtures of the above components ( $N_2$ ,  $CH_4$ ,  $C_2H_6$ ,  $C_3H_8$ , n and  $iC_4H_{10}$ ) and only the saturated liquid between 95 - 150 K was to be considered. Near the end of the project n and  $iC_5H_{12}$  were added to the list of allowable components but no experimental PVTx of pure n and  $iC_5H_{12}$  or binary systems containing n and  $iC_5H_{12}$  were measured as part of the project. The inclusion of these two components is based on data from Orrit, et al. [29] and Orrit, et al. [30].

Four models were considered: the extended corresponding states model, a hard sphere model, a cell model and a Revised Klosek and McKinley model. With the exception of the Revised Klosek and McKinley model only pure fluid and binary system data were used to optimize the models. In the case of the Revised Klosek and McKinley model multicomponent PVTx data were used in the optimization process.

Over a normal range of LNG composition and temperature all four of the models predict densities which agree to within 0.1% of experiment. This is true of all of the experimental PVTx measurements on LNG like mixtures made as part of this project.

No equation or mathematical model based on experimental data can be more accurate than the original data and therefore the extent to which the original goal of 0.1% accuracy has been met depends entirely upon the accuracy of the experimental data referenced above.

There is no reason to doubt the experimental data and therefore there is every reason to believe that the goal of the project has been achieved.

## 2. EXTENDED CORRESPONDING STATES

The thermodynamic equations for the extended corresponding states method are developed in a paper by Rowlinson and Watson [35] and only a very brief description will be given here. Leach [20] developed transformation functions for hydrocarbons which are called shape factor functions. Mollerup [27] and Mollerup and Rowlinson [26] combined the earlier work with the equation of state for methane by Goodwin [8] to produce a computer program to calculate the density of LNG mixtures, which was further modified by Mollerup [28].

The computer program in Appendix F for the calculation of LNG densities based on the extended corresponding states method is an extensive revision of the Mollerup program. Earlier revisions were reported by McCarty [23] and Haynes, Hiza and McCarty [13].

The extended corresponding states method is defined by the following equations:

$$Z_i[P,T] = Z_o[P h_{ii,o}/f_{ii,o}, T/f_{ii,o}] \quad (1)$$

$$G_i[P,T] = f_{ii,o} G_o[P h_{ii,o}/f_{ii,o}, T/f_{ii,o}] - RT \ln(h_{ii,o}) \quad (2)$$

where  $Z$  is the compressibility factor,  $G$  is the Gibbs free energy,  $P$  is pressure and  $T$  is temperature. The subscript  $o$  denotes the reference fluid and the subscript  $i$  denotes the fluid for which properties are to be obtained via the equation of state for the reference fluid and the transformation functions  $f_{ii,o}$  and  $h_{ii,o}$ . The double subscript  $ii$  is introduced now to allow extension of the method to mixtures. The two defining eqs (1) and (2) are necessary since there are two transformation functions. In this case the equation of state for methane by McCarty [22] was chosen for the reference fluid. During the course of the study it was necessary to modify the equation of state by McCarty [22] to give a realistic vapor liquid phase boundary down to a temperature of 43 K. This modification was necessary to accommodate the very low reduced temperatures of the heavier hydrocarbons and was accomplished without changing the performance of the equation of state above the triple point of methane. The equation of state is given in Appendix B.

The  $f_{ii,o}$  and  $h_{ii,o}$  are defined as

$$f_{ii,o} = (T_{ii}^C/T_o^C) \theta_{ii,o}(T_r, V_r) \quad (3)$$

and

$$h_{ii,o} = (V_{ii,o}^C/V_o^C) \phi_{ii,o}(T_r, V_r) \quad (4)$$

where

$$\theta_{ii,o} = 1 + (w_i - w_o)[n_1 - n_2 \ln T_{r_i} + (n_3 - n_4/T_{r_i})(V_{r_i} - n_5)] \quad (5)$$

and

$$\phi_{ii,0} = \frac{Z_0^C}{Z_i^C} \left[ 1 + (w_i - w_0) [n_6(V_{r_i} - n_7) - n_8(V_{r_i} - n_9) \ln T_{r_i}] \right] \quad (6)$$

The  $V_{r_i}$  and  $T_{r_i}$  are reduced temperature and volume, (i.e.,  $T_{r_i} = T/T_{ii}^C$  and  $V_{r_i} = V/V_{ii}^C$ ) each fluid requires a unique  $w_i$  which was estimated using pure fluid experimental data. A single set of the  $n_i$ 's are used for all fluids. The  $n_i$ 's were estimated using all of the pure fluid experimental data from this study. The  $Z_0^C/Z_i^C$  is the ratio of the compressibility factors ( $Z^C = P_c V_c / RT_c$ ) at the critical point. The parameters  $n_j$ ,  $n_j$ ,  $w_i$  and  $Z_{ij}$  are given in Appendix B. All of these parameters were estimated using the experimental PVT data set from this laboratory and least squares estimation techniques.

The extension of the above to mixtures is now accomplished by the following application of the following combining rules:

$$h_{x,0} = \sum_i \sum_j x_i x_j h_{ij,0} \quad (7)$$

$$f_{x,0} h_{x,0} = \sum_i \sum_j x_i x_j f_{ij,0} h_{ij,0} \quad (8)$$

$$f_{ij,0} = \xi_{ij} (f_{ii,0} f_{jj,0})^{1/2} \quad (9)$$

$$h_{ij,0} = \eta_{ij} \left( \frac{1}{2} h_{ii,0}^{1/3} + \frac{1}{2} h_{jj,0}^{1/3} \right)^3 \quad (10)$$

The  $\xi_{ij}$  and the  $\eta_{ij}$  are binary interaction parameters determined by least squares from the PVTx data for binary mixtures. These parameters are given in Appendix B.



This method works quite well as may be seen in the comparisons in Appendix A. It has indeed reproduced all of the present experimental data set to within  $\pm 0.1\%$  except for 14 out of a total of 285 experimental data points. Of these 14 points, 11 are judged to have an uncertainty greater than 0.1%. Figure 1 presents the deviations between the calculated and experimental densities for these 14 points. Appendix A contains comparisons of calculated and experimental densities for the entire data set. This is the best performance of the four models presented here. No pressure, temperature or composition restrictions have been placed on this model.

In the interim publications by McCarty [23] and Haynes, Hiza and McCarty [13] some doubt about the accuracy of the calculated densities was expressed because of the disagreement with a few binary and multicomponent systems containing methane and butane. This disagreement has since been resolved by additional measurements (Haynes [9], Haynes [10] and Miller and Hiza [25] on some of the systems which agree with the predictions of the model but disagree with the previous measurements. The net result of the new measurements is a very slight change in binary interaction coefficients of the methane-butane and nitrogen-butane system. These changes have no practical effect on LNG like mixtures where the concentrations of  $N_2$ ,  $iC_4H_{10}$  and  $nC_4H_{10}$  are individually less than 5%. In other words either the models presented here or those in the interim publications may be used to predict the density of a LNG like mixture to within 0.1% of the true density.

### 3. A HARD SPHERE METHOD

The model of Rodosevich and Miller [33] is one of many modifications of the Longuet-Higgins and Widom [21] model, and was chosen to be included in this study



as a representative example of the application of the hard sphere equation of state concept to the correlation of PVTx data. The equation of state by Rodosevich and Miller [33] is

$$\frac{PV}{RT} = c \frac{1 + y + y^2}{(1 - y)^3} - \frac{a}{RTV} \quad (11)$$

where the  $y = b/4V$  and  $a$ ,  $b$ , and  $c$  are adjustable parameters,  $P$  is pressure,  $V$  is specific volume,  $T$  is temperature and  $R$  is the gas constant. The equation is applied to mixtures by assuming the one-fluid theory and applying the following combining rules.

$$a_m = \sum_i \sum_j a_{ij} x_i x_j \quad (12)$$

$$b_m = \sum_i \sum_j b_{ij} x_i x_j \quad (13)$$

$$c_m = \sum_i \sum_j c_{ij} x_i x_j \quad (14)$$

The mixing rules are:

$$b_{ij} = \left[ \frac{b_{ii}^{1/3} + b_{jj}^{1/3}}{2} (1 - j_{ij}) \right]^3 \quad (15)$$

$$a_{ij} = (a_{ii} a_{jj})^{1/2} \left[ \frac{b_{ij}^2}{b_{ii} b_{jj}} \right]^{1/2} (1 - k_{ij}) \quad (16)$$

$$c_{ij} = \frac{c_{ii} + c_{jj}}{2} \quad (17)$$

The parameters  $j_{ij}$  and  $k_{ij}$  are in this case the binary interaction parameters. The  $a$ 's,  $b$ 's,  $c$ 's,  $j_{ij}$ 's and  $k_{ij}$ 's are given in Appendix B. The excess volume is now calculated using the equation of state and

$$V_E = \tilde{V}_m - \sum_i \tilde{V}_i x_i \quad (18)$$

where  $\tilde{V}_m$  and the  $\tilde{V}_i$  are calculated via the eqs 11 through 17 and then

$$V_m = \sum_i V_i x_i + V_E \quad (19)$$

where the  $V_E$  is from eq (18) and the  $V_i$  are from experimental data. The values of  $V_i$  in this case were calculated from the equations for the liquid density of the pure fluids given in Appendix C.

The above equations are those of Rodosevich and Miller [33] and Rodosevich [34] and only the  $j_{ij}$ 's and  $k_{ij}$ 's have been revised on the basis of the present new data set, and only binary systems data were used to estimate via least squares the  $j_{ij}$ 's and  $k_{ij}$ 's.

As the method is used here it is an excess volume method, and consequently when the temperature of the mixture approaches the critical temperature of one of the component fluids, the method fails. Since the critical temperature of nitrogen is about 126 K, this method should not be used for mixtures containing nitrogen at temperatures above 120 K. Eliminating the data points for mixtures which contain nitrogen at temperatures above 120 K reduces the set from 285 to 251 PVTx points. Figure 2 is a percentage deviation plot containing all of the data points from the set of 251 for which densities calculated by the hard sphere method differ from the experimental density by more than 0.1%. Two things are readily seen in comparing figs. 1 and 2; first, even though total number of points has been reduced in the comparison set, the number of points for which deviations exceed 0.1% in the hard sphere comparison, fig. 2, is far more than for the extended corresponding states comparison, fig. 1. Second, the hard sphere method becomes more uncertain for all mixtures, regardless of components as the temperature exceeds 115 K.

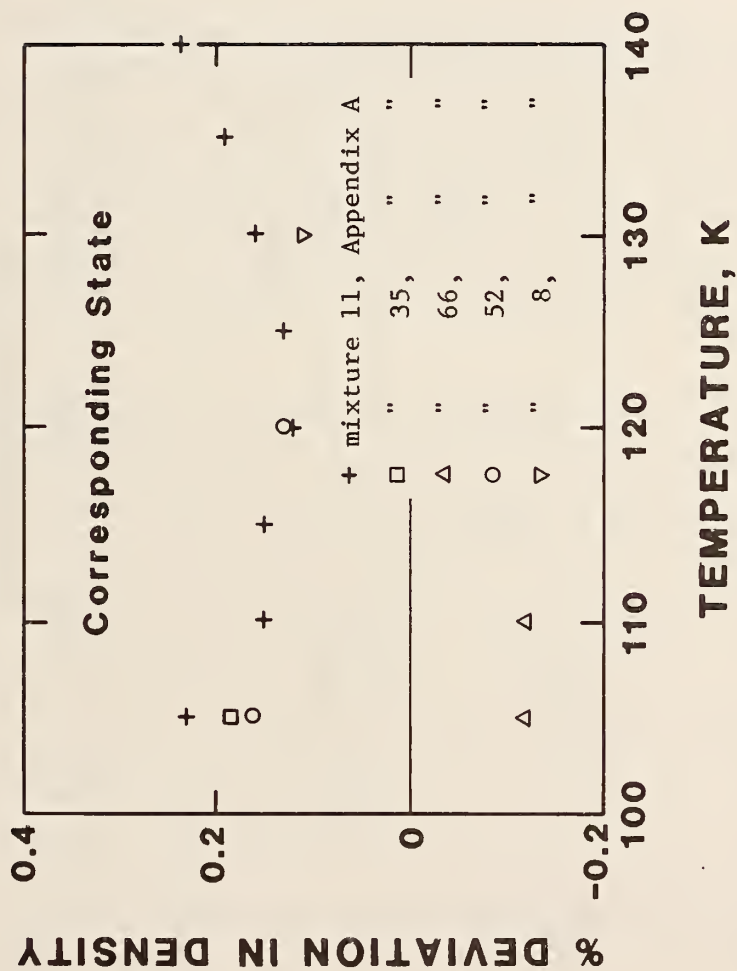


Figure 1. All deviations greater than 0.1% between experimental and calculated densities using the Extended Corresponding States method. The comparison set is all of the data points in Appendix A. (285 data points)

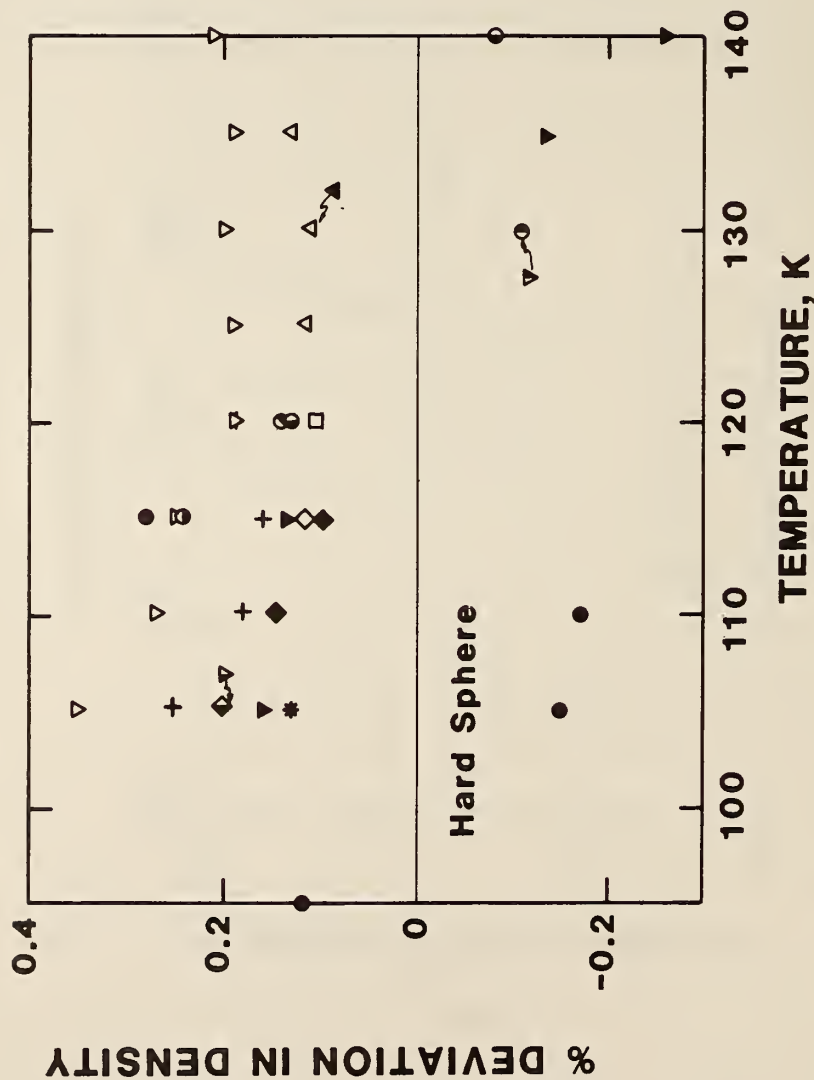


Figure 2. Deviations greater than 0.1% between experimental and calculated densities using the Hard Sphere model. The comparison set is all of the data in Appendix A except those data points for mixtures containing N<sub>2</sub> at temperatures above and including 120 K. (251 data points)

#### 4. A REVISED KLOSEK AND MCKINLEY METHOD

The Klosek and McKinley method [18] is a totally empirical recipe for calculating the density of a LNG-like mixture given the temperature and composition. Pressure is not taken into account. However, this does not seem to be a serious omission. The procedure proposed by Klosek and McKinley [18] is as follows:

$$V_{\text{mix}} = \sum X_i V_i - k X_{\text{CH}_4} \quad (20)$$

where  $V_{\text{mix}}$  is the volume of the mixture,  $X_i$  and  $V_i$  are the mole fraction and volume of the  $i^{\text{th}}$  component,  $X_{\text{CH}_4}$  is the mole fraction of methane and  $k$  is a correction factor obtained from a table or graph. The  $V_i$  and  $k$  are obviously temperature dependent and in addition  $k$  is dependent upon the molecular weight of the mixture.

Using the present data set  $k$  was calculated for all of the experimental data points where methane was present in the mixture and excluding all data points where  $\text{N}_2$  was present in greater than 5% concentration. Figure 3 shows a typical isotherm for  $k$ , with  $\text{N}_2$  present (labeled  $k_2$ ) and without  $\text{N}_2$  present (labeled  $k_1$ ). All of the isotherms available show similar behavior, i.e., all of the  $k$ 's for mixtures containing nitrogen (of about 5%) fall on one line and all of those for mixtures without nitrogen fall on another. Since all of the mixtures with nitrogen have about the same amount of nitrogen present (about 4.5%), the method was modified by adding a term to take into account the nitrogen when it is present. The equation becomes

$$V_{\text{mix}} = \sum X_i V_i - [k_1 + (k_2 - k_1) X_{\text{N}_2} / .0425] X_{\text{CH}_4} \quad (21)$$

where everything is the same as in eq (20) except that  $k_1$  is read from one



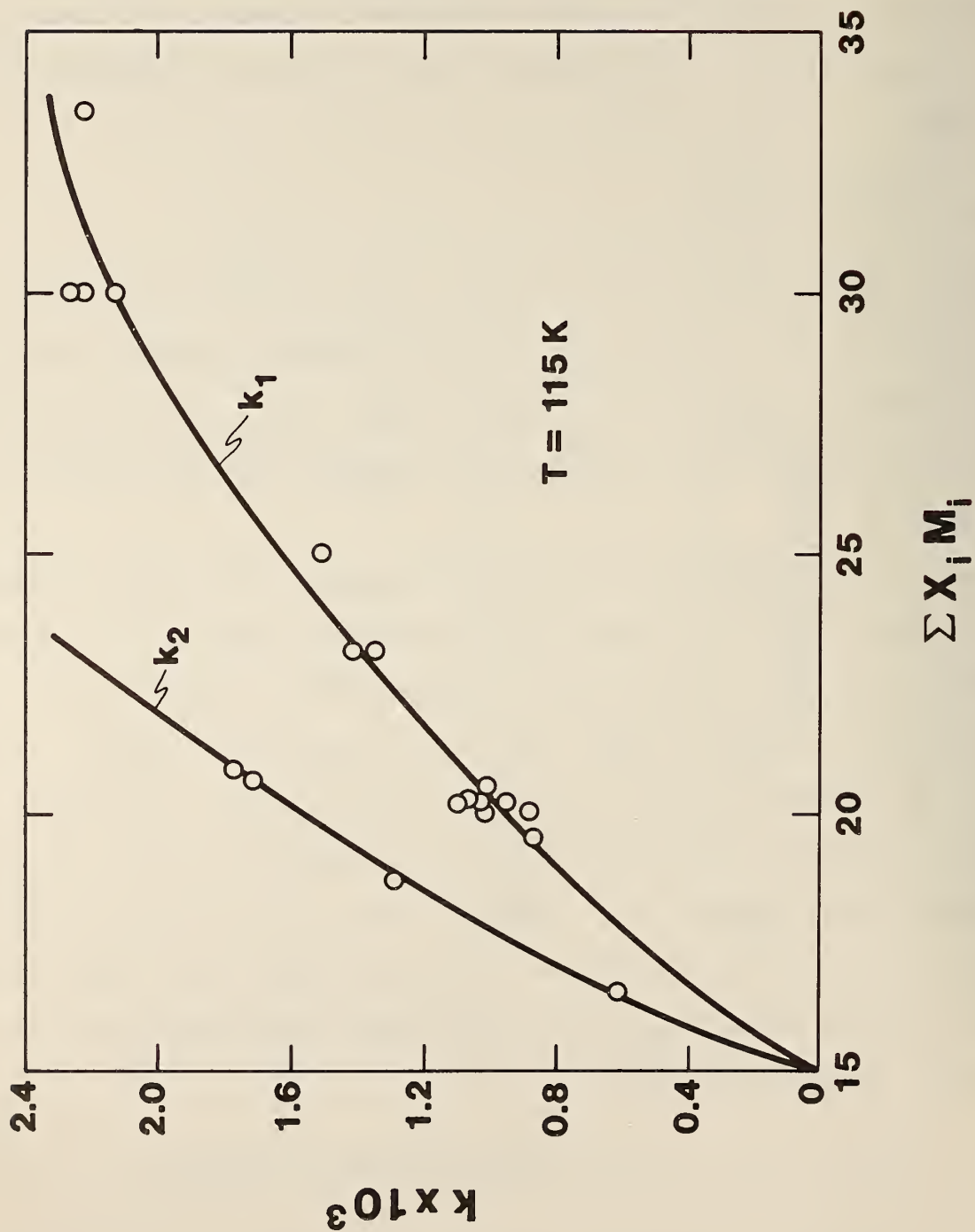


Figure 3. A plot of the correction factors ( $k_1$  and  $k_2$ ) for the 115 kelvin isotherms.

curve and  $k_2$  is read from the other. Appendix D gives tables of values for the  $V_i$ ,  $k_1$  and  $k_2$  which are spaced such that linear interpolation is adequate in both variables (i.e., temperature or molecular weight). The  $k$  factors in Appendix D have been obtained graphically from the multicomponent PVTx data of Hiza and Haynes [15] and Miller and Hiza [25] as well as densities calculated from the extended corresponding states method of section 2.

The limits of compositions of the revised Klosek and McKinley method are the most severe of any of the methods given here. This method should not be used for mixtures other than LNG like mixtures and for LNG like mixtures only when they contain at least 60% methane, less than 4% nitrogen, less than 4% each of  $iC_4H_{10}$   $nC_4H_{10}$  and less than 2% total of  $iC_5H_{12}$  and  $nC_5H_{12}$ .

There are 40 experimental PVTx points from the original set of 285 which may be considered LNG like and fall within the composition limits outlined above. Figure 4 shows all of the deviations between calculated and experimental densities in this 40 point comparison set which exceeds the 0.1% criterion. The deviation trends for the revised Klosek and McKinley method (fig. 4) are very similar to those of the hard sphere method (fig. 2) and in fact all of the deviations in fig. 4 occur at temperatures at or above 115 K, therefore the method can only be considered as accurate as the others for LNG like mixtures at temperatures below 115 K.

## 5. THE CELL MODEL

The cell model considered here was originally proposed by Renon, et al. [32]. In a paper by the same three authors which appeared simultaneously (Eckert, et al. [7]), the cell model was applied to mixtures via Scott's [36] two-fluid theory and a three parameter corresponding states theory. Albright [2] further modified the method by modifying the mixing rules on the basis of a

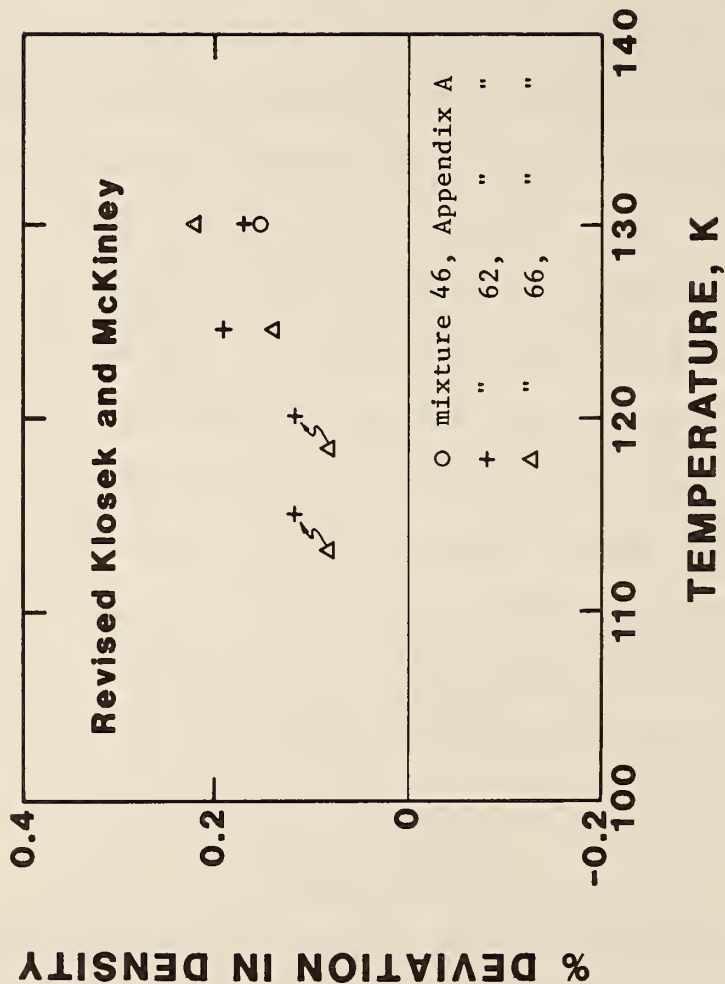


Figure 4. All deviations greater than 0.1% between experimental and calculated densities using the Revised Klosek and McKinley model. The comparison set is all multicomponent mixture data in Appendix A with  $> 60\% \text{CH}_4$ ,  $< 4\% \text{nC}_4\text{H}_{10}$ ,  $< 4\% \text{iC}_4\text{H}_{10}$ ,  $< 4\% \text{N}_2$  and  $< 2\%$  of  $\text{n} + \text{iC}_5\text{H}_{12}$ . (40 data points)

proposal by Yuan [38] and by inserting a pressure dependence based on the experimental liquid ethane data by Pope [31].

The optimization of this method was carried out by M. Albright [1] at Phillips Petroleum Company in Bartlesville, Oklahoma and the details of this work will be published elsewhere. The model is included here because it was optimized to the same data set as the others and therefore the comparisons between experimental and calculated densities given here in fig. 5 together with figs. 1, 2 and 4 provide a common basis of comparison with the other three methods. A listing of the computer program is given in Appendix F.

The same data set as was used in the hard sphere method for comparison has been used here, i.e., all of the data points for mixtures containing nitrogen at temperatures 120 K and above have been taken out of the original 285 points leaving a total of 251 data points.

As in the case of the other methods fig. 5 shows all of the points for which the calculated and experimental densities differ by more than 0.1%.

## 6. USE OF THE METHODS

When the project started in 1972, the atomic weights of nitrogen, carbon and hydrogen were taken from the 1961 carbon 12 scale, IUPAC [16]. During the course of the investigation a revision, Atomic Weights of the Elements [3], to this scale appeared. The revision changed slightly the atomic weights of carbon and hydrogen, but since the changes were small (the maximum difference in any of the densities used here is 0.003%), and because changing the atomic weights would not change the relative results, the changes were not made. Therefore when using the tables and programs in the appendices, the molecular weights given in the tables and programs should be used to maintain consistency.

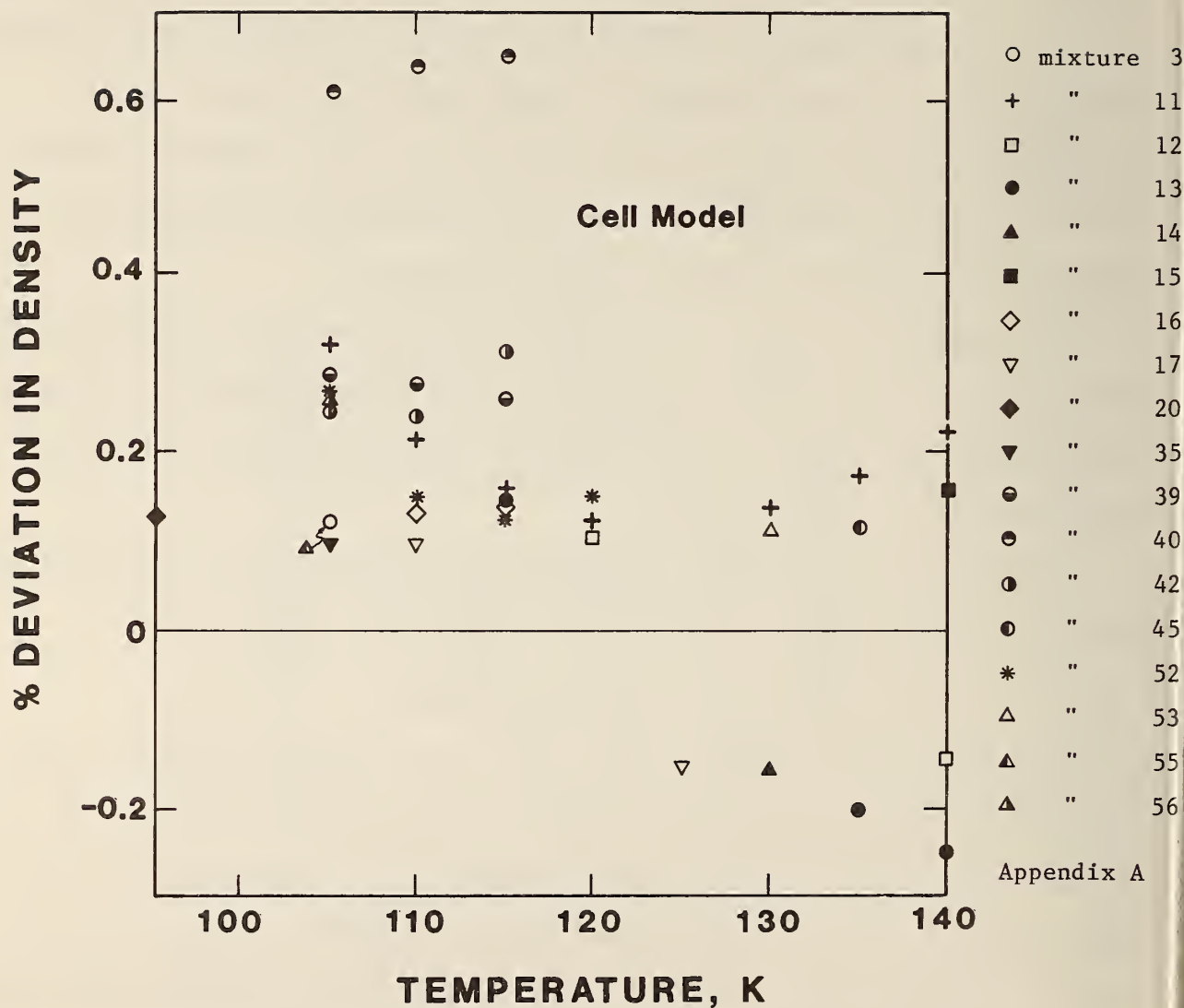


Figure 5. Deviations greater than 0.1% between experimental and calculated densities using the Cell model. The comparison set is all of the data in Appendix A except those data points for mixtures containing  $N_2$  at temperatures above and including 120 K. (251 data points)



The critical parameters used here are from:  $\text{CH}_4$ , McCarty [22];  $\text{C}_2\text{H}_6$ , Sliwinski [37];  $\text{C}_3\text{H}_8$ , Das, et al. [4];  $\text{iC}_4\text{H}_{10}$ , Das, et al. [5];  $\text{nC}_4\text{H}_{10}$ , Das, et al. [6];  $\text{iC}_5\text{H}_{12}$ , Kudchadker, et al. [19]; and  $\text{N}_2$ , Jacobsen, et al. [17].

Errors in the input variables will of course, cause errors in the density predicted by the models. In general, the error in density caused by an error in the input variables is a function of those input variables, and must be treated on an individual basis. However, for LNG like mixtures certain general trends are found. An error in the pressure must be at least 50% before it will have any effect at all on the resulting density. An error in composition, unless it is of the order of several percent, will cause the same relative error in density as it will cause in the molecular weight of the mixture, i.e., if an error in composition causes a 0.1% error in the resulting molecular weight, it will also cause a 0.1% error in the predicted density.

The error in the calculated density due to an error in the input temperature is a function of the composition and the temperature. Table 1 gives resulting errors in density for a 1% error in temperature, for three hypothetical LNG like mixtures.

In general the errors in density caused by an error in temperature are the largest for mixtures containing a high concentration of the most volatile fluids,  $\text{CH}_4$  and  $\text{N}_2$ , and correspondingly the errors decrease as the concentration of the heavier hydrocarbons increases in the mixture. These errors are not a function of which model is being used.

When using the extended corresponding states method, one should keep in mind that twelve significant figures are required by the methane equation of state. The hard sphere model also uses the methane equation from McCarty [22] and the nitrogen equation of Jacobsen, et al. [17] to calculate compressibilities and

TABLE 1. Errors in Density Caused by an Error in the  
Input Temperature of 1%.

Temperature K	% Error in Density		
	Mix A <sup>*</sup>	Mix B <sup>*</sup>	Mix C <sup>*</sup>
95	0.28	0.25	0.20
100	0.30	0.27	0.22
105	0.32	0.29	0.24
110	0.35	0.32	0.29
115	0.39	0.34	0.31

Mix A<sup>\*</sup> = 0.95 CH<sub>4</sub>, 0.05 N<sub>2</sub>

Mix B<sup>\*</sup> = 0.9 CH<sub>4</sub>, 0.02 C<sub>2</sub>H<sub>6</sub>, 0.02 C<sub>3</sub>H<sub>8</sub>, 0.02 iC<sub>4</sub>H<sub>10</sub>, 0.02 nC<sub>4</sub>H<sub>10</sub>, 0.02 N<sub>2</sub>

Mix C<sup>\*</sup> = 0.6 CH<sub>4</sub>, 0.3 C<sub>2</sub>H<sub>6</sub>, 0.02 C<sub>3</sub>H<sub>8</sub>, 0.02 iC<sub>4</sub>H<sub>10</sub>, 0.02 nC<sub>4</sub>H<sub>10</sub>, 0.02 N<sub>2</sub>,  
0.02 iC<sub>5</sub>H<sub>12</sub>.

\*Arbitrary LNG like compositions assumed for the purpose of illustrating the effect of an error in the input temperature.

therefore requires twelve significant figures to insure the accuracy of the calculated density. The other two models require only eight significant figures to be carried along in the calculations.

## 7. CONCLUSIONS

On the basis of the performance of the four models given here and subject to the composition and temperature restrictions already noted, it is estimated that given the pressure, temperature and composition of LNG, any one of the four models may be used to predict the density to within 0.1% of the true value. As has already been mentioned (see section 1) the above accuracy statement is dependent entirely upon the accuracy of the experimental data in Haynes, et al. [11], Haynes, et al. [13], Hiza, et al. [14], Haynes [9], Hiza and Haynes [15], Miller and Hiza [25] and Haynes [10]. These data have been estimated by the authors to be accurate to within 0.1% of the true value with a precision of a few hundredths of a percent. The work on the models given here have provided no basis for questioning the claims of the experimenters, in fact the ability of the models to predict the densities of the multicomponent mixtures to within 0.1% of the measured values tends to support the accuracy claims of the experimenters.

Interim results of this study were reported by Haynes, et al. [13] and McCarty [23], both of which contain earlier versions of the mathematical models given here. These earlier versions are only slightly different than the final ones and for the purposes of calculating LNG densities either of the versions may be used. The reader is, however, cautioned to read the limitations of each model as defined in the earlier sections.

Computer programs for the four models are available at the Thermophysical Properties Division of the National Bureau of Standards in Boulder, Colorado.

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## Appendix A. Experimental Data

The following is a list of all of the experimental PVTx mixtures data which were measured during the course of this project. The data are from Miller and Hiza [25]; Haynes and Hiza [12]; Haynes, et al. [11]; Hiza, et al. [14]; Hiza and Haynes [15] and Haynes [9,10).

All of the data are for the orthobaric liquid except for the data of Miller and Hiza [25] which are for the single phase liquid phase, very close to the orthobaric conditions. The units of the data are bars, moles per liter and kelvin. The columns labeled RKM, HS, CELL and CS correspond to percentage derivations between experimental and predicted densities by the Revised Klosek and McKinley, hard sphere, cell and extended corresponding states models respectively. The derivations are always calculated using experimental-calculated densities.

MIXTURE NO 1 (Miller and Hiza [25])

0.85147 CH<sub>4</sub> + 0.14853 C<sub>2</sub>H<sub>6</sub>

P	D	T	MW	RKM	HS	CELL	CS
.977	25.7499	110.08	18.1265	.13	- .02	.02	.01

MIXTURE NO 2 (Hiza, et al. [14])

0.68006 CH<sub>4</sub> + 0.31994 C<sub>2</sub>H<sub>6</sub>

.416	25.1027	105.00	20.5309	.14	.05	.10	.07
.645	24.7802	110.00	20.5309	.07	- .01	.02	.01
.961	24.4612	115.00	20.5309	.05	- .04	-.03	-.03
1.380	24.1402	120.00	20.5309	.06	- .06	-.06	-.05
1.930	23.8212	125.00	20.5309	.08	- .05	-.04	-.05
2.620	23.5007	130.00	20.5309	.12	- .01	.00	-.02

MIXTURE NO 3 (Hiza, et al. [14])

0.49325 CH<sub>4</sub> + 0.50675 C<sub>2</sub>H<sub>6</sub>

.325	23.9619	105.00	23.1513	.14	.07	.11	.07
.503	23.6937	110.00	23.1513	.09	.03	.03	.03
.749	23.4328	115.00	23.1513	.11	.03	.01	.03
1.080	23.1559	120.00	23.1513	.07	- .02	-.06	-.03
1.500	22.8933	125.00	23.1513	.08	.00	-.05	.00
2.050	22.6290	130.00	23.1513	.14	.04	-.02	.03
2.720	22.3581	135.00	23.1513	.19	.06	.01	.05
3.550	22.0765	140.00	23.1513	.25	.05	.02	.04

MIXTURE NO 4 (Hiza, et al. [14])

0.35457 CH<sub>4</sub> + 0.64543 C<sub>2</sub>H<sub>6</sub>

.256	23.1033	105.00	25.0965	- .04	- .07	-.04	-.07
.397	22.8777	110.00	25.0965	- .01	- .04	-.05	-.05
.580	22.6478	115.00	25.0965	.03	- .02	-.06	-.03
.826	22.4035	120.00	25.0965	.00	- .06	-.11	-.07
1.146	22.1872	125.00	25.0965	.10	.04	-.03	.03
1.550	21.9441	130.00	25.0965	.10	.03	-.04	.02

MIXTURE NO 5 (Hiza, et al. [14])

0.85796 CH<sub>4</sub> + 0.14204 C<sub>3</sub>H<sub>8</sub>

P	D	T	MW	RKM	HS	CELL	CS
.517	24.9622	105.00	20.0279	.01	.10	.10	.07
.817	24.6332	110.00	20.0279	- .02	.08	.07	.06
1.199	24.2942	115.00	20.0279	- .04	.04	.03	.03
1.785	23.9492	120.00	20.0279	- .07	.00	.00	.00
2.415	23.5942	125.00	20.0279	- .09	- .06	-.04	-.05
3.290	23.2461	130.00	20.0279	- .11	- .06	-.01	-.04

MIXTURE NO 6 (Hiza, et al. [14])

0.74920 CH<sub>4</sub> + 0.25080 C<sub>3</sub>H<sub>8</sub>

.478	23.4767	105.00	23.0790	.01	.02	.03	-.02
.738	23.2064	110.00	23.0790	- .03	.00	-.01	-.03
1.099	22.9364	115.00	23.0790	- .01	.00	-.02	-.02
1.582	22.6665	120.00	23.0790	.02	.01	-.01	.01
2.216	22.3818	125.00	23.0790	- .03	- .04	-.04	-.01
3.029	22.1019	130.00	23.0790	.02	- .05	-.02	.01

MIXTURE NO 7 (Hiza, et al. [14])

0.49637 CH<sub>4</sub> + 0.50363 C<sub>3</sub>H<sub>8</sub>

.384	20.4909	105.00	30.1720	- .05	.01	.04	-.03
.591	20.3046	110.00	30.1720	- .08	.01	.01	-.02
.874	20.1180	115.00	30.1720	- .06	.01	-.02	-.02
1.250	19.9311	120.00	30.1720	.00	.01	-.03	.00
1.730	19.7471	125.00	30.1720	.04	.03	-.01	.04
2.320	19.5546	130.00	30.1720	- .01	.01	-.03	.05

MIXTURE NO 8 (Hiza, et al. [14])

0.29538 CH<sub>4</sub> + 0.70462 C<sub>3</sub>H<sub>8</sub>

.271	18.5132	105.00	35.8106	- .19	- .08	-.05	-.12
.409	18.3624	110.00	35.8106	- .24	- .09	-.10	-.12



MIXTURE NO 9 (Haynes [10])

0.92788 CH<sub>4</sub> + 0.07212 nC<sub>4</sub>H<sub>10</sub>

P	D	T	MW	RKM	HS	CELL	CS
1.820	24.2615	120.00	19.0779	.00	.04	-.04	-.04
2.547	23.8868	125.00	19.0779	.00	.00	-.07	-.05
3.470	23.5047	130.00	19.0779	-.01	-.03	-.09	-.07
4.616	23.1225	135.00	19.0779	.04	-.03	-.06	-.05
6.023	22.7284	140.00	19.0779	.16	-.04	-.03	-.03

MIXTURE NO 10 (Haynes [10])

0.92780 CH<sub>4</sub> + 0.07220 nC<sub>4</sub>H<sub>10</sub>

1.270	24.6285	115.00	19.0813	.01	.08	.01	-.01
1.824	24.2783	120.00	19.0813	.07	.11	.03	.04
2.549	23.8999	125.00	19.0813	.06	.07	-.01	.01

MIXTURE NO 11 (Hiza, et al. [14])

0.91674 CH<sub>4</sub> + 0.08326 nC<sub>4</sub>H<sub>10</sub>

.521	25.1536	105.00	19.5467	.21	.35	.33	.22
.810	24.7960	110.00	19.5467	.12	.26	.21	.15
1.216	24.4512	115.00	19.5467	.14	.24	.18	.15
1.753	24.0889	120.00	19.5467	.10	.19	.11	.11
2.472	23.7370	125.00	19.5467	.15	.18	.10	.13
3.374	23.3789	130.00	19.5467	.17	.19	.12	.16
4.509	23.0110	135.00	19.5467	.21	.19	.15	.18
5.887	22.6391	140.00	19.5467	.35	.20	.22	.23

MIXTURE NO 12 (Haynes [10])

0.77982 CH<sub>4</sub> + 0.22018 nC<sub>4</sub>H<sub>10</sub>

1.702	21.6066	120.00	25.3805	.12	.14	.11	.06
2.369	21.3549	125.00	25.3805	.11	.08	.02	.04
3.228	21.1020	130.00	25.3805	.09	.00	-.07	.03
4.291	20.8555	135.00	25.3805	.12	-.03	-.10	.07
5.576	20.5978	140.00	25.3805	.20	-.13	-.15	.07

## MIXTURE NO 13 (Haynes [10])

0.77762 CH<sub>4</sub> + 0.22238 nC<sub>4</sub>H<sub>10</sub>

P	D	L	T	MW	RKM	HS	CEL	CS
1.179	21.8054	115.00	25.4011	.07	.15	.16	.02	
1.699	21.5601	120.00	25.4011	.07	.10	.07	.02	
2.372	21.3164	125.00	25.4011	.10	.06	.01	.03	
3.230	21.0605	130.00	25.4011	.06	- .03	-.10	.00	
4.291	20.8011	135.00	25.4011	.02	- .14	-.20	-.03	
5.579	20.5448	140.00	25.4011	.09	- .23	-.26	-.03	

## MIXTURE NO 14 (Hiza, et al. [14])

0.58828 CH<sub>4</sub> + 0.41172 nC<sub>4</sub>H<sub>10</sub>

3.183	18.3058	130.00	33.3687	- .32	- .11	-.15	-.08	
2.342	18.4853	125.00	33.3687	- .17	.00	.00	-.03	
2.281	18.4772	125.00	33.3687	- .21	- .05	-.04	-.08	
1.636	18.6495	120.00	33.3687	- .18	.02	.07	-.06	

## MIXTURE NO 15 (Haynes [10])

0.92044 CH<sub>4</sub> + 0.07956 iC<sub>4</sub>H<sub>10</sub>

1.254	24.3633	115.00	19.3910	- .21	.05	.06	-.01	
1.805	24.0029	120.00	19.3910	- .24	.02	.03	-.03	
2.521	23.6403	125.00	19.3910	- .22	.01	.04	-.03	
3.434	23.2752	130.00	19.3910	- .21	.03	.08	.00	
4.567	22.8920	135.00	19.3910	- .22	.00	.09	-.02	
5.950	22.5037	140.00	19.3910	- .14	- .02	.14	-.02	

## MIXTURE NO 16 (Haynes [10])

0.78329 CH<sub>4</sub> + 0.21671 iC<sub>4</sub>H<sub>10</sub>

.782	21.9144	110.00	25.1625	- .33	.10	.15	.01	
1.164	21.6652	115.00	25.1625	- .33	.05	.07	-.01	
1.671	21.4136	120.00	25.1625	- .35	.01	.00	-.02	
2.329	21.1668	125.00	25.1625	- .34	.00	-.03	.00	
3.170	20.9125	130.00	25.1625	- .35	- .04	-.07	.00	
4.208	20.6629	135.00	25.1625	- .33	- .06	-.06	.04	
5.474	20.4082	140.00	25.1625	- .23	- .09	-.04	.07	

MIXTURE NO 17 (Hiza, et al. [14])

0.48687 CH<sub>4</sub> + 0.51313 iC<sub>4</sub>H<sub>10</sub>

P	D	T	MW	RKM	HS	CELL	CS
.629	17.3575	110.00	37.6362	- .62	.05	.11	.02
.938	17.2076	115.00	37.6362	- .70	.00	-.01	-.02
1.361	17.0639	120.00	37.6362	- .64	-.02	-.08	-.02
1.852	16.9156	125.00	37.6362	- .69	-.07	-.17	-.04

MIXTURE NO 18 (Hiza, et al. [14])

0.95248 CH<sub>4</sub> + 0.04752 N<sub>2</sub>

1.380	26.8476	105.00	16.6119	- .05	-.04	-.04	-.02
1.990	26.4052	110.00	16.6119	.06	.00	.00	.04
2.634	25.9374	115.00	16.6119	.20	.01	.02	.06
3.500	25.4522	120.00	16.6119	.26	*	.03	.07
4.600	24.9496	125.00	16.6119	.54	*	.04	.07
5.830	24.4210	130.00	16.6119	.69	-5.69	.03	.05
7.300	23.8600	135.00	16.6119	.77	-3.99	-.01	-.03
9.200	23.2809	140.00	16.6119	1.09	-2.69	-.02	-.07

MIXTURE NO 19 (Hiza, et al. [14])

0.69651 CH<sub>4</sub> + 0.30349 N<sub>2</sub>

3.450	26.8735	100.00	19.6759	- .39	.06	.09	.00
4.661	26.3393	105.00	19.6759	- .56	.02	.07	.04
6.181	25.7686	110.00	19.6759	- .92	-.05	-.02	.00
8.010	25.1790	115.00	19.6759	-1.28	.13	-.08	-.03
10.112	24.5737	120.00	19.6759	-1.56	1.88	-.07	-.04

MIXTURE NO 20 (Hiza, et al. [14])

0.50758 CH<sub>4</sub> + 0.49242 N<sub>2</sub>

3.303	27.0801	95.00	21.9375	-2.16	.11	.14	.06
4.468	26.4588	100.00	21.9375	-1.92	-.01	.05	.04
6.383	25.8106	105.00	21.9375	-1.56	-.15	-.05	-.01
8.461	25.1387	110.00	21.9375	-1.29	-.17	-.09	-.03
10.740	24.4431	115.00	21.9375	1.12	.28	-.04	-.01
13.983	23.7096	120.00	21.9375	-1.79	3.41	.07	-.01
17.529	22.9315	125.00	21.9375	-1.14	*	.26	.00
21.076	22.1005	130.00	21.9375	-.72	-2.84	.05	.10

\* The hard sphere solution for the density of N<sub>2</sub> failed.

## MIXTURE NO 21 (Hiza, et al. [14])

0.67287 C<sub>2</sub>H<sub>6</sub> + 0.32713 C<sub>3</sub>H<sub>8</sub>

P	D	T	MW	RKM	HS	CELL	CS
	18.6192	125.00	34.6588	.00	.02	-.02	.00
	18.4648	130.00	34.6588	.00	.02	.00	.01
	18.3059	135.00	34.6588	.00	.00	.00	.00
	18.1509	140.00	34.6588	.00	.01	.04	.02

## MIXTURE NO 22 (Hiza, et al. [14])

0.50105 C<sub>2</sub>H<sub>6</sub> + 0.49895 C<sub>3</sub>H<sub>8</sub>

	18.3618	105.00	37.0689		.02	.04	.02
	18.2169	110.00	37.0689		.01	.00	.00
	18.0726	115.00	37.0689		.00	-.03	-.01
	17.9282	120.00	37.0689	-	.01	-.05	-.01
	17.7880	125.00	37.0689		.00	-.03	.00
	17.6412	130.00	37.0689	-	.02	-.04	-.02
	17.4988	135.00	37.0689	-	.01	-.01	-.01
	17.3526	140.00	37.0689	-	.03	.00	-.01

## MIXTURE NO 23 (Hiza, et al. [14])

0.67117 C<sub>2</sub>H<sub>6</sub> + 0.32883 nC<sub>4</sub>H<sub>10</sub>

	17.5047	110.00	39.2952		.00	.02	.00
	17.3706	115.00	39.2952		.01	.00	.00
	17.1031	125.00	39.2952		.02	.00	.02
	16.9626	130.00	39.2952	-	.01	-.03	-.02
	16.8285	135.00	39.2952		.00	-.01	-.01
	16.6947	140.00	39.2952		.00	.03	.01

## MIXTURE NO 24 (Hiza, et al. [14])

0.65343 C<sub>2</sub>H<sub>6</sub> + 0.34657 nC<sub>4</sub>H<sub>10</sub>

	17.2184	115.00	39.7929		.00	.00	.00
	17.0824	120.00	39.7929	-	.02	-.03	-.02

## MIXTURE NO 25 (Hiza, et al. [14])

0.72436 C<sub>2</sub>H<sub>6</sub> + 0.27564 iC<sub>4</sub>H<sub>10</sub>

	17.9779	105.00	37.8030		.01	.01	.01
	17.8401	110.00	37.8030		.01	.00	.01
	17.4235	125.00	37.8030		.01	.00	.01
	17.2825	130.00	37.8030		.00	.01	.01

MIXTURE NO 26 (Hiza, et al. [14])

0.68939 C<sub>2</sub>H<sub>6</sub> + 0.31061 iC<sub>4</sub>H<sub>10</sub>

P	D	T	MW	RKM	HS	CELL	CS
	17.3716	115.00	38.7840		.00	-.03	.00
	17.2344	120.00	38.7840		- .02	-.04	-.02

MIXTURE NO 27 (Hiza, et al. [14])

0.94067 C<sub>2</sub>H<sub>6</sub> + 0.05933 N<sub>2</sub>

3.850	21.4718	105.00	29.9481		- .02	-.01	-.03
4.630	21.2912	110.00	29.9481		.03	.02	.03
5.472	21.0845	115.00	29.9481		- .01	-.04	-.04
6.383	20.8998	120.00	29.9481		- .72	.01	.00

MIXTURE NO 28 (Hiza, et al. [14])

0.60949 C<sub>3</sub>H<sub>8</sub> + 0.39051 nC<sub>4</sub>H<sub>10</sub>

	14.6487	115.00	49.5749		- .01	-.02	-.01
	14.5521	120.00	49.5749		.02	.00	.02

MIXTURE NO 29 (Hiza, et al. [14])

0.60650 C<sub>3</sub>H<sub>8</sub> + 0.39350 nC<sub>4</sub>H<sub>10</sub>

	14.1343	140.00	49.6169		.00	-.03	.00
	14.0333	145.00	49.6169		.00	.00	.00
	13.9346	150.00	49.6169		.02	.05	.02

MIXTURE NO 30 (Hiza, et al. [14])

0.58692 C<sub>3</sub>H<sub>8</sub> + 0.41308 nC<sub>4</sub>H<sub>10</sub>

	14.6839	110.00	49.8915		.01	.04	.01
	14.1748	135.00	49.8915		- .03	-.06	-.03
	14.0786	140.00	49.8915		.00	-.02	.00

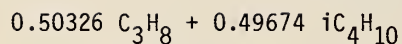
MIXTURE NO 31 (Hiza, et al., [14])

0.49030 C<sub>3</sub>H<sub>8</sub> + 0.50970 iC<sub>4</sub>H<sub>10</sub>

	14.3080	105.00	51.2468		- .01	-.04	-.01
	14.2136	110.00	51.2468		- .01	-.05	.00
	14.1219	115.00	51.2468		.02	-.03	.02
	14.0257	120.00	51.2468		.01	-.04	.01
	13.9300	125.00	51.2468		.00	-.04	.00
	13.8342	130.00	51.2468		- .01	-.04	-.01

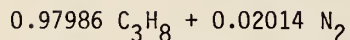


MIXTURE NO 32 (Hiza, et al. [14])



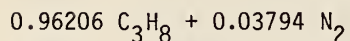
P	D	T	MW	RKM	HS	CELL	CS
	13.9718	125.00	51.0650		.01	-.03	.01
	13.8737	130.00	51.0650		- .01	-.04	-.01

MIXTURE NO 33 (Hiza, et al. [14])



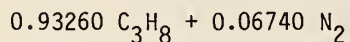
3.567	16.2131	110.00	43.7733		- .01	-.04	-.01
4.712	16.0931	115.00	43.7733		- .02	-.05	-.02

MIXTURE NO 34 (Hiza, et al. [14])



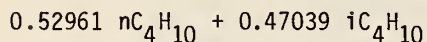
4.955	16.4638	105.00	43.4870		.02	-.03	.01
6.708	16.3410	110.00	43.4870		.00	-.04	.00

MIXTURE NO 35 (Hiza, et al. [14])



8.795	16.7084	105.00	43.0132		.20	.11	.18
6.313	16.8055	100.00	43.0132		.04	-.07	.01

MIXTURE NO 36 (Hiza, et al. [14])



12.6943	125.00	58.1243		- .01	-.03	-.01
12.6133	130.00	58.1243		.01	-.01	.01
12.5271	135.00	58.1243		- .01	-.04	-.01
12.4447	140.00	58.1243		.00	-.02	.00

MIXTURE NO 37 (Hiza and Haynes [15])



.476	24.9975	105.00	20.1852	.02	.04	.06	.02
.747	24.6696	110.00	20.1852	- .03	.00	.00	-.02
1.119	24.3515	115.00	20.1852	.01	.02	.02	.01
1.616	24.0335	120.00	20.1852	.07	.07	.07	.06

MIXTURE NO 38 (Hiza and Haynes [15])



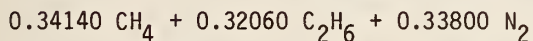
P	D	T	MW	RKM	HS	CELL	CS
.427	20.5342	110.00	30.0903	- .03	.02	.00	-.03
.636	20.3497	115.00	30.0903	- .01	.03	-.02	-.02
.638	20.3567	115.00	30.0903	.03	.06	.01	.01
.914	20.1786	120.00	30.0903	.12	.10	.04	.06
.914	20.1803	120.00	30.0903	.12	.11	.05	.07
1.280	19.9858	125.00	30.0903	.10	.08	.01	.04

MIXTURE NO 39 (Hiza and Haynes [15])



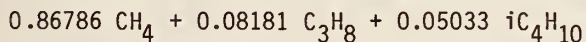
3.688	25.8587	105.00	20.3301	.57	.01	.28	.00
4.813	25.4598	110.00	20.3301	.64	- .03	.27	-.01
6.181	25.0466	115.00	20.3301	.79	.02	.25	-.05
7.731	24.6359	120.00	20.3301	.70	.42	.31	-.04

MIXTURE NO 40 (Hiza and Haynes [15])



7.190	24.9084	105.00	24.5861	2.06	.01	.61	.07
9.494	24.5350	110.00	24.5861	2.55	- .01	.64	.04
12.372	24.1433	115.00	24.5861	3.72	.24	.65	-.06

MIXTURE NO 41 (Miller and Hiza [25])



.466	24.9239	100.00	20.4561	- .17	.04	.11	-.03
.769	24.4200	108.00	20.4561	- .17	.01	.05	-.04
1.327	23.9676	115.00	20.4561	- .18	- .01	.02	-.05

MIXTURE NO 42 (Hiza and Haynes [15])



2.817	25.3950	105.00	20.1181	.17	.08	.24	-.01
3.620	25.0153	110.00	20.1181	.10	.02	.23	-.05
4.549	24.6465	115.00	20.1181	.08	.08	.33	.01
5.610	24.2667	120.00	20.1181	.02	*	.44	.04

\* The hard sphere solution for the density of N<sub>2</sub> failed.

MIXTURE NO 43 (Haynes [9])

0.89071 CH<sub>4</sub> + 0.04998 nC<sub>4</sub>H<sub>10</sub> + 0.05931 N<sub>2</sub>

P	D	T	MW	RKM	HS	CELL	CS
2.400	25.3450	110.00	18.8562	- .23	.00	.01	.03
3.145	24.9440	115.00	18.8562	- .25	- .04	-.02	.01
4.082	24.5383	120.00	18.8562	- .38	*	-.02	.00
5.196	24.1141	125.00	18.8562	- .32	*	-.04	-.04

MIXTURE NO 44 (Miller and Hiza [25])

0.85317 CH<sub>4</sub> + 0.05077 C<sub>2</sub>H<sub>6</sub> + 0.04855 C<sub>3</sub>H<sub>8</sub> + 0.04751 nC<sub>4</sub>H<sub>10</sub>

1.038	24.6056	110.04	20.1165	.02	.15	.10	.06
1.464	24.2712	115.00	20.1165	.00	.11	.04	.02

MIXTURE NO 45 (Haynes [9])

0.85133 CH<sub>4</sub> + 0.05759 C<sub>2</sub>H<sub>6</sub> + 0.04808 C<sub>3</sub>H<sub>8</sub> + 0.04300 nC<sub>4</sub>H<sub>10</sub>

1.180	24.3243	115.00	20.0092	- .04	.07	.00	-.01
1.700	23.9965	120.00	20.0092	.00	.10	.02	.03
2.374	23.6586	125.00	20.0092	.05	.11	.04	.05
3.232	23.3108	130.00	20.0092	.04	.11	.05	.05
4.301	22.9634	135.00	20.0092	.08	.13	.11	.09

MIXTURE NO 46 (Haynes [9])

0.84566 CH<sub>4</sub> + 0.07924 C<sub>2</sub>H<sub>6</sub> + 0.05060 C<sub>3</sub>H<sub>8</sub> + 0.02450 nC<sub>4</sub>H<sub>10</sub>

1.167	24.5569	115.00	19.6051	.00	.04	.00	-.01
1.683	24.2126	120.00	19.6051	.02	.04	-.01	-.01
2.350	23.8698	125.00	19.6051	.08	.07	.03	.03
3.201	23.5204	130.00	19.6051	.12	.11	.09	.07

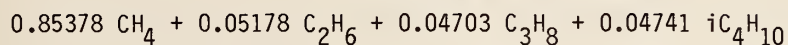
MIXTURE NO 47 (Haynes [9])

0.86040 CH<sub>4</sub> + 0.04600 C<sub>2</sub>H<sub>6</sub> + 0.04790 C<sub>3</sub>H<sub>8</sub> + 0.04570 iC<sub>4</sub>H<sub>10</sub>

1.186	24.2654	115.00	19.9552	- .14	.01	.04	-.03
1.710	23.9371	120.00	19.9552	- .10	.04	.07	.01
2.387	23.5860	125.00	19.9552	- .10	.01	.05	-.01
3.248	23.2331	130.00	19.9552	- .13	- .01	.06	-.02
4.320	22.8637	135.00	19.9552	- .18	- .07	.05	-.07

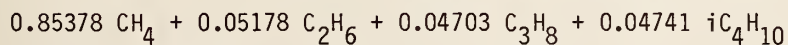
\* The hard sphere solution for the density of N<sub>2</sub> failed.

MIXTURE NO 48 (Haynes [9])



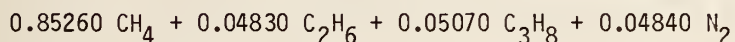
P	D	T	MW	RKM	HS	CELL	CS
1.190	24.2100	115.00	20.0838	- .14	.01	.04	-.03
1.706	23.8779	120.00	20.0838	- .14	.01	.04	-.02
2.379	23.5324	125.00	20.0838	- .13	- .02	.02	-.04
3.238	23.1834	130.00	20.0838	- .15	- .04	.03	-.05

MIXTURE NO 49 (Miller and Hiza [25])



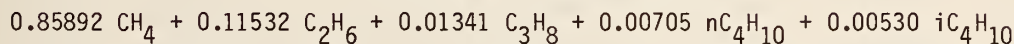
.972	24.5434	110.02	20.0838	- .14	.03	.06	-.01
1.429	24.2154	115.01	20.0838	- .12	.02	.05	-.01

MIXTURE NO 50 (Hiza and Haynes [15])



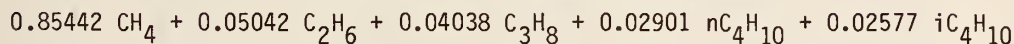
1.581	25.8910	105.00	18.7223	.01	- .04	.03	-.07
2.138	25.5081	110.00	18.7223	- .09	- .07	.03	-.07
2.827	25.1224	115.00	18.7223	- .05	- .06	.05	-.06
3.650	24.7283	120.00	18.7223	- .08	*	.08	-.06

MIXTURE NO 51 (Haynes [9])



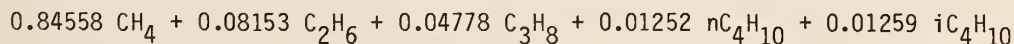
1.185	25.0957	115.00	18.5565	.05	- .02	-.01	-.03
1.706	24.7131	120.00	18.5565	.03	- .06	-.05	-.07
2.372	24.3294	125.00	18.5565	.02	- .07	-.05	-.09
3.225	23.9490	130.00	18.5565	.05	- .02	.01	-.06

MIXTURE NO 52 (Hiza and Haynes [15])



.515	24.8775	105.00	20.1885	.09	.25	.25	.16
.818	24.5382	110.00	20.1885	.00	.17	.15	.09
1.190	24.2083	115.00	20.1885	.00	.15	.12	.08
1.695	23.8859	120.00	20.1885	.05	.19	.15	.13

MIXTURE NO 53 (Haynes [9])



1.166	24.5586	115.00	19.5838	.01	.05	.04	.01
1.680	24.2180	120.00	19.5838	.04	.07	.05	.03
2.348	23.8688	125.00	19.5838	.08	.08	.08	.04
3.188	23.5154	130.00	19.5838	.10	.10	.12	.07

\* The hard sphere solution for the density of N<sub>2</sub> failed.

MIXTURE NO 54 (Haynes [9])

0.81249 CH<sub>4</sub> + 0.08484 C<sub>2</sub>H<sub>6</sub> + 0.04931 C<sub>3</sub>H<sub>8</sub> + 0.02708 nC<sub>4</sub>H<sub>10</sub> + 0.02628 N<sub>2</sub>

P	D	T	MW	RKM	HS	CELL	CS
2.214	24.4562	115.00	20.0706	.01	.01	.03	-.02
2.874	24.1119	120.00	20.0706	.00	*	.04	-.01
3.768	23.7507	125.00	20.0706	.02	*	.02	-.05
4.793	23.3954	130.00	20.0706	- .02	-3.67	.07	-.03

MIXTURE NO 55 (Hiza and Haynes [15])

0.79090 CH<sub>4</sub> + 0.05600 C<sub>2</sub>H<sub>6</sub> + 0.05000 C<sub>3</sub>H<sub>8</sub> + 0.04770 nC<sub>4</sub>H<sub>10</sub> + 0.05540 N<sub>2</sub>

1.933	24.8080	105.00	20.9017	.04	.10	.12	.05
2.530	24.4664	110.00	20.9017	.02	.02	.03	-.02

MIXTURE NO 56 (Miller and Hiza [25])

0.79054 CH<sub>4</sub> + 0.05597 C<sub>2</sub>H<sub>6</sub> + 0.04996 C<sub>3</sub>H<sub>8</sub> + 0.04762 nC<sub>4</sub>H<sub>10</sub> + 0.05591 N<sub>2</sub>

2.158	24.8354	105.03	20.9029	.15	.21	.23	.15
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MIXTURE NO 57 (Haynes [9])

0.80940 CH<sub>4</sub> + 0.04542 C<sub>2</sub>H<sub>6</sub> + 0.05050 C<sub>3</sub>H<sub>8</sub> + 0.04667 iC<sub>4</sub>H<sub>10</sub> + 0.04801 N<sub>2</sub>

3.005	24.1487	115.00	20.6355	.02	-.02	-.03	-.04
3.863	23.8075	120.00	20.6355	- .14	*	.00	-.03
4.874	23.4518	125.00	20.6355	- .12	*	.01	-.05
6.125	23.0893	130.00	20.6355	- .18	-5.06	.05	-.08

MIXTURE NO 58 (Hiza and Haynes [15])

0.80600 CH<sub>4</sub> + 0.04680 C<sub>2</sub>H<sub>6</sub> + 0.04820 C<sub>3</sub>H<sub>8</sub> + 0.05000 iC<sub>4</sub>H<sub>10</sub> + 0.04900 N<sub>2</sub>

2.039	24.7803	105.00	20.7423	- .03	.08	.03	.01
2.550	24.4327	110.00	20.7423	- .09	-.02	-.06	-.06
3.135	24.1039	115.00	20.7423	.06	.00	-.02	-.02
3.790	23.7707	120.00	20.7423	- .10	*	-.01	-.02

MIXTURE NO 59 (Miller and Hiza [25])

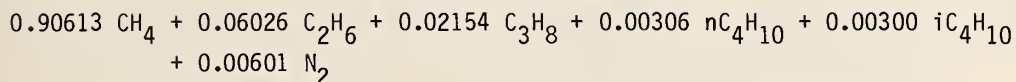
0.80545 CH<sub>4</sub> + 0.04671 C<sub>2</sub>H<sub>6</sub> + 0.04817 C<sub>3</sub>H<sub>8</sub> + 0.04998 iC<sub>4</sub>H<sub>10</sub> + 0.04969 N<sub>2</sub>

2.273	24.7831	105.06	20.7476	.00	.10	.05	.03
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\* The hard sphere solution for the density of N<sub>2</sub> failed.

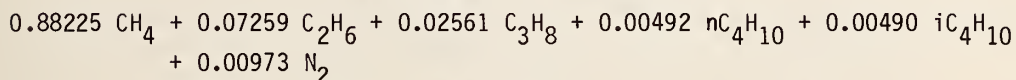


## MIXTURE NO 60 (Haynes [9])



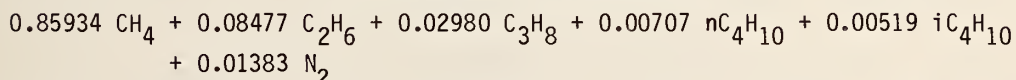
P	D	T	MW	RKM	HS	CELL	CS
1.478	25.3834	115.00	17.8195	.03	- .01	.01	-.01
2.043	24.9894	120.00	17.8195	.08	.00	.05	.02
2.785	24.5702	125.00	17.8195	.04	.00	.03	-.02
3.722	24.1578	130.00	17.8195	.06	-1.08	.11	.02

## MIXTURE NO 61 (Haynes [9])



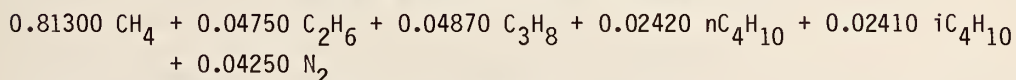
1.639	25.2023	115.00	18.3094	.09	.04	.07	.04
2.247	24.8047	120.00	18.3094	.06	.00	.04	.00
3.022	24.4022	125.00	18.3094	.04	.00	.03	-.03

## MIXTURE NO 62 (Haynes [9])



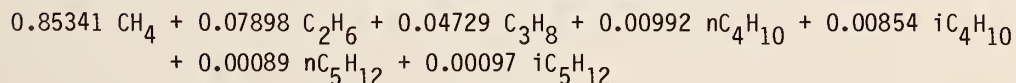
1.812	25.0384	115.00	18.7496	.12	.06	.09	.06
2.441	24.6661	120.00	18.7496	.12	*	.11	.07
3.223	24.2880	125.00	18.7496	.19	.00	.16	.09
4.222	23.8981	130.00	18.7496	.17	-2.12	.20	.10

## MIXTURE NO 63 (Hiza and Haynes [15])



1.834	24.8496	105.00	20.6168	.03	.12	.10	.05
2.807	24.5159	110.00	20.6168	.02	.08	.06	.03
3.384	24.1783	115.00	20.6168	.09	.06	.05	.03
6.039	23.8577	120.00	20.6168	.04	*	.11	.08

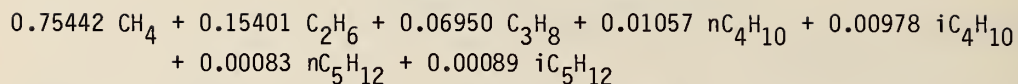
## MIXTURE NO 64 (Haynes [9])



.787	25.0063	110.00	19.3588	.01	.02	.02	-.01
1.172	24.6566	115.00	19.3588	.00	- .01	-.01	-.03
1.686	24.3079	120.00	19.3588	.02	.00	.00	-.02
2.351	23.9525	125.00	19.3588	.05	.01	.03	-.01
3.210	23.5883	130.00	19.3588	.06	.02	.06	.00

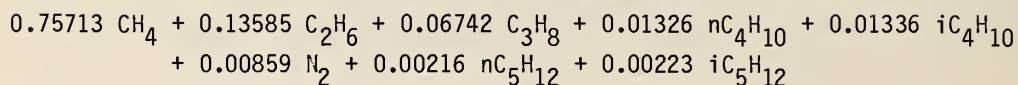
\* The hard sphere solution for the density of N<sub>2</sub> failed.

MIXTURE NO 65 (Haynes [9])



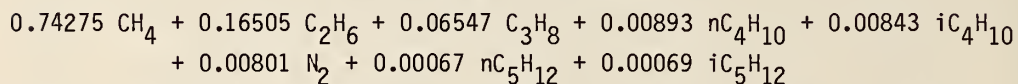
P	D	T	MW	RKM	HS	CELL	CS
.723	24.2529	110.00	21.1060	- .02	.00	-.01	-.04
1.081	23.9619	115.00	21.1060	.04	.04	.03	.01
1.549	23.6535	120.00	21.1060	.07	.04	.02	.01
2.153	23.3351	125.00	21.1060	.04	.01	.00	-.01

MIXTURE NO 66 (Haynes [9])



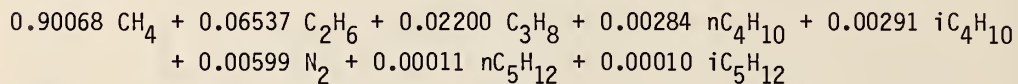
1.155	24.1809	110.00	21.3094	.09	.07	.08	.04
1.595	23.8731	115.00	21.3094	.12	.05	.05	.02
2.155	23.5709	120.00	21.3094	.12	.00	.08	.05
2.873	23.2644	125.00	21.3094	.15	.00	.12	.08
3.744	22.9514	130.00	21.3094	.22	-1.42	.16	.11

MIXTURE NO 67 (Haynes [9])



1.158	24.3141	110.00	21.0976	.02	- .02	.00	-.05
1.584	24.0160	115.00	21.0976	.09	.01	.02	-.01
2.093	23.6937	120.00	21.0976	.03	.00	-.03	-.06
2.853	23.3804	125.00	21.0976	.05	.00	-.01	-.05

MIXTURE NO 68 (Haynes [9])



1.456	25.3600	115.00	17.9026	.05	.01	.03	.01
2.024	24.9656	120.00	17.9026	.08	.00	.06	.03
2.762	24.5450	125.00	17.9026	.03	.00	.03	-.03
3.698	24.1289	130.00	17.9026	.02	-1.13	.07	-.02

## Appendix B. Computer Program and Equation Parameters for the Extended Corresponding States Model

The program listings in Appendix F include the extended corresponding states method described in section 2. To use the program in its present form one must make one of the two possible calls to SUBROUTINE PDMIX(P,D,T,X) (lines LNG 1 through LNG 9). The two possible calls are:

CALL PDMIX(P,D,T,X)

or

CALL PMIX(P,D,T,X)

When the call to PDMIX(P,D,T,X) is made the input variables are: P (pressure in bars); T (temperature in kelvin); and X which is a matrix of the mole fraction of the components of the mixture in the following order:

- X(1) = mole fraction of methane
- X(2) = mole fraction of ethane
- X(3) = mole fraction of propane
- X(4) = mole fraction of normal butane
- X(5) = mole fraction of isobutane
- X(6) = mole fraction of nitrogen
- X(7) = mole fraction of normal pentane
- X(8) = mole fraction of isopentane

No other components are allowed and if one or more of the above components are absent, a zero should be inserted in the appropriate matrix element. The program then calculates a density and it is returned in the argument list as D (density in moles/liter).

When a call to PMIX(P,D,T,X) is made all of the above are the same except that the roles of P and D are interchanged, i.e., D is an input variable and P is calculated by the program.

The range of the program is 90 to 150 kelvin for the saturated liquid phase of any of the pure components of  $\text{CH}_4$ ,  $\text{C}_2\text{H}_6$ ,  $\text{C}_3\text{H}_8$ ,  $\text{nC}_4\text{H}_{10}$ ,  $\text{iC}_4\text{H}_{10}$ ,  $\text{N}_2$ ,  $\text{nC}_4\text{H}_{12}$  or  $\text{iC}_4\text{H}_{12}$  or any mixture of those fluids. The program will extrapolate to higher pressures (higher than saturation pressure) but the user is reminded that such a calculation is an extrapolation and should be used with caution.

Other subprograms required:

SUBROUTINE MIX DATA, line LNG 95, Appendix F

SUBROUTINE DATA CH4, line LNG 162, Appendix F

FUNCTION FINDM, line LNG 217, Appendix F

FUNCTION SATL, line LNG 236, Appendix F

SUBROUTINE PROPS, line LNG 248, Appendix F

The equation of state from which  $Z_0$  and  $G_0$  by eqs (1) and (2) may be derived is:

$$\begin{aligned}
P = & \rho RT + \rho^2(N_1T + N_2T^{1/2} + N_3 + N_4/T + N_5/T^2) \\
& + \rho^3(N_6T + N_7 + N_8/T + N_9/T^2) \\
& + \rho^4(N_{10}T + N_{11} + N_{12}/T) + \rho^5(N_{13}) \\
& + \rho^6(N_{14}/T + N_{15}/T^2) + \rho^7(N_{16}/T) \\
& + \rho^8(N_{17}/T + N_{18}/T^2) + \rho^9(N_{19}/T^2) \\
& + \rho^3(N_{20}/T^2 + N_{21}/T^3) \exp(-\gamma\rho^2) \\
& + \rho^5(N_{22}/T^2 + N_{23}/T^4) \exp(-\gamma\rho^2) \\
& + \rho^7(N_{24}/T^2 + N_{25}/T^3) \exp(-\gamma\rho^2) \\
& + \rho^9(N_{26}/T^2 + N_{27}/T^4) \exp(-\gamma\rho^2) \\
& + \rho^{11}(N_{28}/T^2 + N_{29}/T^3) \exp(-\gamma\rho^2) \\
& + \rho^{13}(N_{30}/T^2 + N_{31}/T^3 + N_{32}/T^4) \exp(-\gamma\rho^2)
\end{aligned} \tag{B-1}$$

The computer subroutine PROPS(PP,DD,TT), lines "LNG 248 through LNG 371 in Appendix F" are the FORTRAN statements for eq (B-1) and the derivative  $(\partial P/\partial \rho)_T$ .

The parameters for eq (B-1) as applied to methane are given in table 2 and in the FORTRAN SUBROUTINE DATA CH4, lines LNG 162 through LNG 216, Appendix F.

The parameters for eqs (5) and (6), (9) and (10) are given in table 3 and 4 and in the FORTRAN SUBROUTINE MIX DATA (IBASE), lines LNG 95 through LNG 161, Appendix F.



Table 2. Methane Coefficients  $N_i$  for Eq B-1.

$R$	$=$	$0.08205616$	$N_{16}$	$=$	$-0.529609525984 \times 10^{-3}$
$\gamma$	$=$	$0.0096$	$N_{17}$	$=$	$0.152264286004 \times 10^{-4}$
$N_1$	$=$	$-0.187027997685 \times 10^{-1}$	$N_{18}$	$=$	$-0.109952182842 \times 10^{-1}$
$N_2$	$=$	$0.103387108009 \times 10$	$N_{19}$	$=$	$0.191395549929 \times 10^{-3}$
$N_3$	$=$	$-0.155387625619 \times 10^2$	$N_{20}$	$=$	$0.386470003746 \times 10^5$
$N_4$	$=$	$0.772311478564 \times 10^3$	$N_{21}$	$=$	$-0.157930582612 \times 10^7$
$N_5$	$=$	$-0.377103300895 \times 10^5$	$N_{22}$	$=$	$0.195270144401 \times 10^3$
$N_6$	$=$	$0.846818843475 \times 10^{-3}$	$N_{23}$	$=$	$0.165996081629 \times 10^7$
$N_7$	$=$	$-0.496415884529$	$N_{24}$	$=$	$0.603051146711$
$N_8$	$=$	$0.869909352414 \times 10^2$	$N_{25}$	$=$	$0.376485162808 \times 10^2$
$N_9$	$=$	$-0.322821592493 \times 10^5$	$N_{26}$	$=$	$0.125593680622 \times 10^{-2}$
$N_{10}$	$=$	$-0.395843026318 \times 10^{-4}$	$N_{27}$	$=$	$-0.343570032513 \times 10^2$
$N_{11}$	$=$	$0.266772318035 \times 10^{-1}$	$N_{28}$	$=$	$-0.540945094139 \times 10^{-5}$
$N_{12}$	$=$	$-0.304010057839 \times 10$	$N_{29}$	$=$	$0.185622284663 \times 10^{-2}$
$N_{13}$	$=$	$0.191584507536 \times 10^{-3}$	$N_{30}$	$=$	$0.770786979245 \times 10^{-8}$
$N_{14}$	$=$	$-0.195587933458 \times 10^{-3}$	$N_{31}$	$=$	$-0.286868318650 \times 10^{-5}$
$N_{15}$	$=$	$0.607479967879 \times 10$	$N_{32}$	$=$	$0.372376961647 \times 10^{-4}$

Table 3. Coefficients for Eqs 5 and 6.

$n_1$	=	-0.109495	$n_4$	=	-4.14192
$n_2$	=	0.919454	$n_5$	=	0.444850
$n_3$	=	-4.01525			
$n_6$	=	0.356808	$n_8$	=	0.893323
$n_7$	=	1.02619	$n_9$	=	0.761533

		$P_c^*$	$T_c$	$V_c$		Fluid
	w	(bar)	(K)	(cm <sup>3</sup> /mol)	M	No.
CH <sub>4</sub>	0.0109	45.956967	190.555	98.522	16.04303	1
C <sub>2</sub> H <sub>6</sub>	0.110427	48.60314	305.5	146.2	30.07012	2
C <sub>3</sub> H <sub>8</sub>	0.154837	42.445123	370.	200.	44.09721	3
nC <sub>4</sub> H <sub>10</sub>	0.176372	38.295398	425.	251.62	58.1243	4
iC <sub>4</sub> H <sub>10</sub>	0.150115	36.88998	408.1	263.	58.1243	5
N <sub>2</sub>	0.0291791	33.542557	126.2	89.827	28.0134	6
nC <sub>5</sub> H <sub>12</sub>	0.234320	33.812152	469.6	304.	72.15139	7
iC <sub>5</sub> H <sub>12</sub>	0.288886	31.988302	460.39	306.	72.15139	8

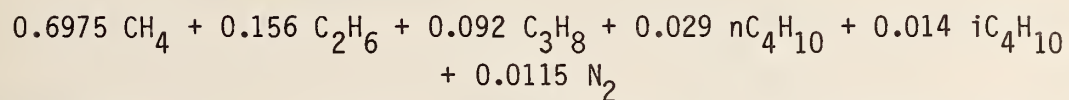
\*Note: The large number of significant figures given for critical pressure is necessary to reproduce the  $Z_c$  ( $Z_c = P_c V_c / RT_c$ ) in the least squares fit of the data.  $P_c$ 's have been converted to bar from atmospheres.

Table 4. Binary Interaction Coefficients for Eqs 9 and 10.  
Fluid numbers are given in table 3.

i	ij							
	1	2	3	4	5	6	7	8
1	1	1.00514	1.01922	1.04243	1.05048	1.01009	1.06	1.06
2		1	1.00599	1.01616	1.02369	1.02127	1.02	1.02
3			1	1.00172	1.01140	1.04606	1.01	1.01
4				1	0.997114	1.20955	1.0	1.0
5					1	1.13889	1.0	1.0
6						1	1.0	1.0
7							1	1.0
8								1

i	ij								
	j	1	2	3	4	5	6	7	8
1	1	1	1.01127	0.988608	0.983130	0.986978	0.953430	0.98	0.98
2			1	0.999961	0.972223	0.998886	0.939622	0.99	0.99
3				1	0.985547	1.03099	0.912209	0.99	0.99
4					1	0.976416	0.849200	0.99	0.99
5						1	0.857310	0.99	0.99
6							1	0.99	0.99
7								1	1.0
8									1

Table 5. Values for Checking Calculations Using Corresponding States Equations.\*



Temperature K	Density in moles/liter	Pressure, bar
95	24.333	1
100	24.067	1.1
105	23.796	1.2

\* Included for check purposes only; these values are calculated from the corresponding states model and are not experimental data.

## Hard Sphere Model

The program listing that follows is for the hard sphere model described in section 3. To use the program in its present form one must make the following reference to the computer program:

$$\text{DEN} = \text{RODEN}(\text{P}, \text{T}, \text{X})$$

where DEN is density in moles/liter, P is pressure in bars, T is temperature in kelvin, and X is a matrix of the mole fractions of the components of the mixture in the following order:

- X(1) = mole fraction of methane
- X(2) = mole fraction of ethane
- X(3) = mole fraction of propane
- X(4) = mole fraction of normal butane
- X(5) = mole fraction of isobutane
- X(6) = mole fraction of nitrogen
- X(7) = mole fraction of normal pentane
- X(8) = mole fraction of isopentane

Note: the inclusion of the pentanes is due to Rodosevich and Miller [33] and no optimization of parameters has been included in this work for mixtures with pentane as a component.

The range of the program is 90 to 150 kelvin for the saturated liquid phase of mixtures of  $\text{CH}_4$ ,  $\text{C}_2\text{H}_6$ ,  $\text{C}_3\text{H}_8$ ,  $n\text{C}_4\text{H}_{10}$ ,  $i\text{C}_4\text{H}_{10}$ ,  $\text{N}_2$ ,  $n\text{C}_5\text{H}_{12}$  and  $i\text{C}_5\text{H}_{12}$ . The program will calculate densities of any of the pure components but they will be from a different model (i.e., some from an equation of state ( $\text{CH}_4$  and  $\text{N}_2$ ) and some from the equations for saturated liquid densities. Therefore in its present form, extrapolation to higher pressures is possible but the reliability of the results is questionable.



Other subprograms required:

SUBROUTINE FM, lines LNG 435 to 529, Appendix F

SUBROUTINE ZERO, lines LNG 617 to 623, Appendix F

FUNCTION FIND V1, lines LNG 530 to 548, Appendix F

FUNCTION FIND G1, lines LNG 549 to 560, Appendix F

FUNCTION EXCESS, lines LNG 561 to 616, Appendix F

FUNCTION FIND M, lines LNG 217 to 234, Appendix F

SUBROUTINE PROPS, lines LNG 248 to 371, Appendix F

SUBROUTINE DATA CH4, lines LNG 162 to 216, Appendix F

SUBROUTINE DATA N2, lines LNG 372 to 415, Appendix F

The parameters for eq (11), section 3 are given in table 6, and in lines LNG 462 through LNG 466 in Appendix F.

The binary interaction parameters  $j_{ij}$  and  $k_{ij}$  in eqs (15) and (16), section 3 are given in table 7 and in lines LNG 447 through LNG 461 in Appendix F.

Table 6. Coefficients for Eq 11.

Fluid	$a_i$	$S_i$	$c_i$	Fluid No.
$\text{CH}_4$	$2.755 \times 10^5$	$3.676 \times 10^{-8}$	1.00	1
$\text{C}_2\text{H}_6$	$7.773 \times 10^5$	$4.158 \times 10^{-8}$	1.50	2
$\text{C}_3\text{H}_8$	$14.165 \times 10^5$	$4.644 \times 10^{-8}$	1.67	3
$\text{nC}_4\text{H}_{10}$	$22.733 \times 10^5$	$5.051 \times 10^{-8}$	1.83	4
$\text{iC}_4\text{H}_{10}$	$21.279 \times 10^5$	$5.056 \times 10^{-8}$	1.79	5
$\text{N}_2$	$1.718 \times 10^5$	$3.546 \times 10^{-8}$	1.03	6
$\text{nC}_5\text{H}_{12}$	$30.550 \times 10^5$	$5.389 \times 10^{-8}$	1.91	7
$\text{iC}_5\text{H}_{12}$	$42.946 \times 10^5$	$5.706 \times 10^{-8}$	2.11	8

$$b_i = (2)(3.14159)(6.025 \times 10^{23})S_i^3/3$$

Table 7. Binary Interaction Parameters for Eqs 15 and 16.

		$j_{ij}$							
$i$	$j$	1	2	3	4	5	6	7	8
1	0	-0.388616	$\times 10^{-2}$	$-0.120932 \times 10^{-1}$	$-0.231577 \times 10^{-1}$	$-0.238349 \times 10^{-1}$	$-0.997547 \times 10^{-2}$	$-0.326 \times 10^{-1}$	$0.458 \times 10^{-1}$
2	0		0	$-0.2162 \times 10^{-2}$	$-0.400910 \times 10^{-2}$	$-0.812712 \times 10^{-2}$	$-0.143976 \times 10^{-1}$	$-0.3 \times 10^{-2}$	$-0.4 \times 10^{-2}$
3				0	$0.761571 \times 10^{-3}$	$-0.383743 \times 10^{-2}$	$-0.24014 \times 10^{-1}$	0.0	0.0
4					0	$0.222150 \times 10^{-2}$	$-0.576043 \times 10^{-1}$	0.0	0.0
5						0	$-0.576043 \times 10^{-1}$	0.0	0.0
6							0	$0.4 \times 10^{-1}$	$0.5 \times 10^{-1}$
7								0.0	0.0
8									0.0

		$k_{ij}$							
$i$	$j$	1	2	3	4	5	6	7	8
1	0	$0.298830 \times 10^{-2}$		$0.597378 \times 10^{-1}$	0.110893	0.100298	$0.197290 \times 10^{-1}$	0.14	0.1745
2	0		0	$0.14527 \times 10^{-1}$	$0.67703 \times 10^{-1}$	$0.346632 \times 10^{-1}$	$0.529034 \times 10^{-1}$	$0.2 \times 10^{-1}$	$0.3 \times 10^{-1}$
3				0	$0.249291 \times 10^{-1}$	$-0.838212 \times 10^{-2}$	0.14719	0.0	0.0
4					0	$0.199213 \times 10^{-1}$	0.154365	0.0	0.0
5						0	0.154365	0.0	0.0
6							0	0.15	0.18
7									0.0

## Appendix D. Computer Program and Parameters for the Revised

### Klosek and McKinley Model

The program listing and tables that follow are for the Revised Klosek and McKinley model described in section 4. The method may be used in two ways. First using the equation:

$$V_{\text{mix}} = \sum X_i V_i - [k_1 + (k_2 - k_1) X_{N_2} / 0.0425] X_{CH_4} \quad (D-1)$$

The  $V_i$ ,  $k_1$  and  $k_2$  may be obtained from tables 8, 9 and 10 and the volume of the mixture calculated. For example given the mixture of 0.8130  $CH_4$  + 0.0475  $C_2H_6$  + 0.0487  $C_3H_8$  + 0.0242  $nC_4H_{10}$  + 0.0241  $iC_4H_{10}$  + 0.0425  $N_2$  and a temperature of 105 kelvin.

The  $\sum X_i V_i$  and  $\sum X_i W_i$  are obtained from table 8.

$$\begin{aligned} \sum X_i V_i &= (.8130)(.037113) + (.0475)(.047267) + (.0487)(.061766) \\ &\quad + (.0242)(.076100) + (.0241)(.077538) + (.0425)(.042565) \\ &= 0.0409453 \end{aligned}$$

$$\begin{aligned} \text{Then the } \sum X_i W_i &= (.8130)(16.04303) + (.0475)(30.07012) + (.0487)(44.09721) \\ &\quad + (.0242)(58.1243) + (.0241)(58.1243) + (.0425)(28.0143) \\ &= 20.6168 \text{ the molecular weight of the mixture} \end{aligned}$$

$$\text{from table 9 } k_1 = .697 \times 10^{-3}$$

$$\text{from table 10 } k_2 = .849 \times 10^{-3}$$

plugging all this into eq (D-1) gives

$$V_{\text{mix}} = .040255$$

$$1/V_{\text{mix}} = \rho_{\text{mix}} = 24.842 \text{ moles/liter}$$

This compares to the experimental value of 24.850 (Appendix A, mixture No. 63) to within 0.03%.

The same result may be obtained by using the computer program in the following way:

$$D = \text{FMKM}(T,X)$$

where  $T$  is temperature in kelvin and  $X$  is a matrix of mole fractions of the components in the following order:

- $X(1)$  = mole fraction of methane
- $X(2)$  = mole fraction of ethane
- $X(3)$  = mole fraction of propane
- $X(4)$  = mole fraction of normal butane
- $X(5)$  = mole fraction of isobutane
- $X(6)$  = mole fraction of nitrogen
- $X(7)$  = mole fraction of normal pentane
- $X(8)$  = mole fraction of isopentane

for the example:  $T = 105.$ ,  $X(1) = .8130$ ,  $X(2) = .0475$ ,  $X(3) = .0487$ ,  
 $X(4) = .0242$ ,  $X(5) = .0241$  and  $X(6) = .0425$ .

Other subprograms required:

FUNCTION VIDEAL, lines LNG 709 through 732, Appendix F

FUNCTION SAT, lines LNG 733 through 762, Appendix F



Table 8. Volumes of Saturated Liquid of the Pure Components in  
Liters/mole.

T, K	CH <sub>4</sub>	C <sub>2</sub> H <sub>6</sub>	C <sub>3</sub> H <sub>8</sub>	nC <sub>4</sub> H <sub>10</sub>	iC <sub>4</sub> H <sub>10</sub>	N <sub>2</sub>	nC <sub>5</sub> H <sub>12</sub>	iC <sub>5</sub> H <sub>12</sub>
90.	.035441	.046081	.060461	.074708	.076084	.037543	.089173	.089243
92.	.035649	.046235	.060632	.074891	.076274	.038081	.089379	.089454
94.	.035861	.046390	.060804	.075075	.076466	.038650	.089586	.089666
96.	.036077	.046547	.060977	.075259	.076659	.039254	.089793	.089878
98.	.036298	.046704	.061151	.075445	.076853	.039897	.090000	.090091
100.	.036524	.046863	.061325	.075631	.077047	.040586	.090208	.090304
102.	.036755	.047023	.061501	.075818	.077243	.041327	.090416	.090518
104.	.036992	.047185	.061677	.076006	.077440	.042128	.090624	.090733
106.	.037234	.047348	.061855	.076194	.077637	.043002	.090833	.090948
108.	.037481	.047512	.062033	.076384	.077836	.043963	.091042	.091163
110.	.037735	.047678	.062212	.076574	.078035	.045031	.091252	.091379
112.	.037995	.047845	.062392	.076765	.078236	.046231	.091462	.091596
114.	.038262	.048014	.062574	.076957	.078438	.047602	.091673	.091814
116.	.038536	.048184	.062756	.077150	.078640	.049179	.091884	.092032
118.	.038817	.048356	.062939	.077344	.078844	.050885	.092095	.092251
120.	.039106	.048529	.063124	.077539	.079049	.052714	.092307	.092470
122.	.039404	.048704	.063309	.077734	.079255	.054679	.092520	.092690
124.	.039710	.048881	.063496	.077931	.079462	.056797	.092733	.092911
126.	.040025	.049059	.063684	.078128	.079671	.059085	.092947	.093133
128.	.040350	.049239	.063873	.078327	.079880	.061565	.093161	.093355
130.	.040685	.049421	.064063	.078526	.080091	.064263	.093376	.093578
molecular weight*	16.04303	30.07012	44.09721	58.1243	58.1243	28.0143	72.15139	72.15139

Table 9. Correction Factor  $k_1 \times 10^3$ .

T/W	16	17	18	19	20	21	22	23	24	25
90	-.005	.120	.220	.340	.430	.515	.595	.660	.725	.795
95	-.006	.135	.260	.380	.500	.590	.665	.740	.810	.885
100	-.007	.150	.300	.425	.575	.675	.755	.830	.910	.990
105	-.007	.165	.340	.475	.635	.735	.840	.920	1.045	1.120
110	-.008	.180	.375	.535	.725	.835	.950	1.055	1.155	1.245
115	-.009	.220	.440	.610	.810	.945	1.065	1.180	1.280	1.380
120	-.01	.250	.500	.695	.920	1.055	1.205	1.330	1.450	1.550
125	-.013	.295	.590	.795	1.035	1.210	1.385	1.525	1.640	1.750
130	-.015	.345	.700	.920	1.200	1.370	1.555	1.715	1.860	1.990
135	-.017	.400	.825	1.060	1.390	1.590	1.800	1.950	2.105	2.272

Table 10. Correction Factor  $k_2 \times 10^3$ .

T/W	16	17	18	19	20	21	22	23	24	25
90	-.004	.10	.22	.35	.50	.60	.69	.78	.86	.95
95	-.005	.12	.28	.43	.59	.71	.83	.94	1.05	1.14
100	-.007	.16	.34	.49	.64	.79	.94	1.08	1.17	1.27
105	-.01	.24	.42	.61	.75	.91	1.05	1.19	1.33	1.45
110	-.015	.32	.59	.77	.92	1.07	1.22	1.37	1.52	1.71
115	-.024	.41	.72	.95	1.15	1.22	1.3	1.45	1.65	2.00
120	-.032	.60	.91	1.23	1.43	1.63	1.85	2.08	2.30	2.45
125	-.043	.71	1.13	1.48	1.73	1.98	2.23	2.48	2.75	2.90
130	-.058	.95	1.46	1.92	2.20	2.42	2.68	3.00	3.32	3.52
135	-.075	1.30	2.00	2.40	2.60	3.00	3.40	3.77	3.99	4.23

## Appendix E. Computer Program for the Cell Model

The program listings for the cell model start at line LNG 763 and continue on to the end of Appendix F. As is mentioned in section 4, no details of the model are given here only the program listing. To use the program in its present form one must make the following reference to the computer program:

CALL ECKNON(P,D,T,X)

where P is input pressure in bars, D is the output density in moles/liter, T is the input temperature in kelvins and X is a matrix of the mole fractions of the components of the mixture in the following order:

- X(1) = mole fraction of methane
- X(2) = mole fraction of ethane
- X(3) = mole fraction of propane
- X(4) = mole fraction of normal butane
- X(5) = mole fraction of isobutane
- X(6) = mole fraction of nitrogen
- X(7) = mole fraction of normal pentane
- X(8) = mole fraction of isopentane

## Appendix F. Computer Programs

Listing of computer programs for all four models. See the sections on the individual models for a list of subprograms needed for each model.

The programs are written in FORTRAN IV and are operational on a CDC 6600 computer.

	SUBROUTINE PDMIX(P,D,T,X)	LNG	1
C	FOR A CALL TO PDMIX, P,T AND X ARE INPUT. P IS IN BAR,T IS IN	LNG	2
C	KELVIN AND D IS OUTPUT IN THE UNITS OF MOLES/LITER	LNG	3
C	FOR A CALL TO P MIX, D,T AND X ARE INPUT AND P IS OUTPUT, THE	LNG	4
C	UNITS ARE THE SAME	LNG	5
C	THE X MATRIX MUST CONTAIN THE MOL FRACTION OF THE ALLOWABLE FLUIDS	LNG	6
C	IN THE FOLLOWING ORDER,1=C1,2=C2,3=C3,4=NC4,5=IC4,6=N2,7=NC5,8=IC5	LNG	7
C	PLACE A ZERO IN THE ELEMENTS OF X WHERE THAT PARTICULAR GAS IS NOT	LNG	8
C	PRESENT	LNG	9
	DIMENSION ZATA(10,10),ATA(10,10),TC(10),VC(10),ZC(10),AC(10),W(10)	LNG	10
	1,PC(10),CF(9)	LNG	11
	DIMENSION THETA(10,10),TH(10,10),PHI(10,10),PH(10,10),F(10,10),FH	LNG	12
	1(10,10),H(10,10),HH(10,10),VR(10,10),TR(10,10),X(10)	LNG	13
	COMMON/DATA M/ZATA,ATA,TC,VC,W,TCO,VCO,ACO,ZCO,RR,R,OMEGO,AC,ZC,N	LNG	14
	1,PC,CF	LNG	15
	DATA(IE=0)	LNG	16
	D=0.0 \$ PI=P/1.01325	LNG	17
	GO TO 4	LNG	18
	ENTRY PMIX	LNG	19
	P=0.0	LNG	20
	IP=1	LNG	21
	GO TO 5	LNG	22
4	IP=0	LNG	23
5	CONTINUE	LNG	24
	IF(IE.GT.0)GO TO 6	LNG	25
	IE=1	LNG	26
	CALL DATA CH4	LNG	27
	IBASE=L=1	LNG	28
	CALL MIX DATA(IBASE)	LNG	29
6	CONTINUE	LNG	30
	DO 1 I=1,N	LNG	31
	F(I,I)=H(I,I)=1.	LNG	32
	THETA(I,I)=1.	LNG	33
1	PHI(I,I)=1.	LNG	34
	DO 30 J=1,30	LNG	35
	HX=FXHX=0.0	LNG	36
	DO 10 I=1,N	LNG	37
	FH(I,I)=F(I,I)	LNG	38
	HH(I,I)=H(I,I)	LNG	39
	IF(X(I).LT..0001)GO T O 10	LNG	40
	F(I,I)=(TC(I)/TC(L))*THETA(I,I)	LNG	41
	H(I,I)=(VC(I)/VC(L))*PHI(I,I)	LNG	42
10	CONTINUE	LNG	43
	DO 11 IA=1,N	LNG	44
	DO 11 IB=1,N	LNG	45
	IF(X(IA).LT..0001)GO TO 11	LNG	46
	IF(X(IB).LT..0001)GO TO 11	LNG	47
	FAB=ZATA(IA,IB)*(F(IA,IA)* F(IB,IB))**.5	LNG	48
	HAB=ATA(IA,IB)*(.5*H(IA,IA)**(1./3.)+.5*H(IB,IB)**(1./3.))**3	LNG	49
	HX=HX+X(IA)*X(IB)*HAB	LNG	50
	FXHX=FXHX+X(IA)*X(IB)*HAB*FAB	LNG	51
11	CONTINUE	LNG	52
	FX=FXHX/HX	LNG	53
	PRO=PI*HX/FX	LNG	54
	TRO=T/FX	LNG	55
	DEN=D*HX	LNG	56
	IF(IP.EQ.1)GO TO 8	LNG	57
	DD=SATL(TRO)*1000.+1.	LNG	58
9	DEN=FIND M(PRO,TRO,DD)	LNG	59
8	CONTINUE	LNG	60
	IF(DEN.LE.0.0) GO TO 33	LNG	61
	VR0=1000./DEN	LNG	62
	DO 12 I=1,N	LNG	63



IF(X(I).LT..0001)GO TO 12	LNG 64
VR(I,I)=VRO*PHI(I,I)/VCO	LNG 65
TR(I,I)=TRO*THETA(I,I)/TCO	LNG 66
IF(VR(I,I).GT.2.)VR(I,I)=2.	LNG 67
IF(VR(I,I).LT..5)VR(I,I)=.5	LNG 68
TH(I,I)=THETA(I,I)	LNG 69
THETA(I,I)=1.+(AC(I)-OMEGO)*(CF(1)-CF(2)*ALOG(TR(I,I))+(CF(3)-CF(4	LNG 70
1)/TR(I,I))*(VR(I,I)-CF(5)))	LNG 71
PH(I,I)=PHI(I,I)	LNG 72
PHI(I,I)=(1.+(AC(I)-OMEGO)*(CF(6)*(VR(I,I)-CF(7))-CF(8)*(VR(I,I)	LNG 73
1-CF(9))*ALOG(TR(I,I))))*ZCO/ZC(I)	LNG 74
12 CONTINUE	LNG 75
DO 13 I=1,N	LNG 76
IF(X(I).LT..0001)GO TO 13	LNG 77
IF(ABS ((FH(I,I)-F(I,I))/F(I,I)).GT..001)GO TO 30	LNG 78
IF(ABS ((HH(I,I)-H(I,I))/H(I,I)).GT..001)GO TO 30	LNG 79
IF(ABS ((TH(I,I)-THETA(I,I))/THETA(I,I)).GT..001)GO TO 30	LNG 80
IF(ABS ((PH(I,I)-PHI(I,I))/PHI(I,I)).GT..001)GO TO 30	LNG 81
13 CONTINUE	LNG 82
GO TO 31	LNG 83
30 CONTINUE	LNG 84
33 PRINT 100,P,DEN,T	LNG 85
100 FORMAT(* ITTERATION FAILED AT*,3F10.4)	LNG 86
STOP	LNG 87
31 D=DEN/HX	LNG 88
THC=THETA(6,6)	LNG 89
IF(IP.EQ.0)GO TO 32	LNG 90
CALL PRESS(P,DEN,TRO)	LNG 91
PI=1.01325*P*FX/HX	LNG 92
32 RETURN	LNG 93
END	LNG 94
SUBROUTINE MIX DATA(IBASE)	LNG 95
DIMENSION ZATA(10,10),ATA(10,10),TC(10),VC(10),ZC(10),AC(10),W(10)	LNG 96
1,PC(10),CF(9)	LNG 97
COMMON/DATA M/ZATA,ATA,TC,VC,W,TCO,VCO,ACO,ZCO,RR,R,OMEGO,AC,ZC,N	LNG 98
1,PC,CF	LNG 99
DATA(PC=45.356,47.96757,41.89008,37.79462,36.40758,33.10393)	LNG 100
DATA(TC=190.555,305.5,370.,425.,408.1,126.2)	LNG 101
DATA(W=16.04303,30.07012,44.09721,58.1243,58.1243,28.0134)	LNG 102
DATA(VC=98.522,146.2,200.,251.62,263.00,89.827)	LNG 103
DATA(AC=.0109,.110427,.154837,.176372,.150115,.0291791)	LNG 104
DATA(R=82.05606),(RR=8.3144),(N=6)	LNG 105
DATA(CF=-.109495,.919454,-4.01525,-4.14192,.444850,.356808,	LNG 106
11.02619,.893323,.761533)	LNG 107
DATA(ATA(1,2)=1.00514),(ATA(1,3)=1.01922),(ATA(1,4)=1.04243),	LNG 108
1(ATA(1,5)=1.05048),(ATA(1,6)=1.01009),	LNG 109
A(ATA(2,3)=1.00599),(ATA(2,4)=1.01616),	LNG 110
2(ATA(2,5)=1.02369),(ATA(2,6)=1.02127),(ATA(3,4)=1.00172),	LNG 111
3(ATA(3,5)=1.01140),(ATA(3,6)=1.04606),(ATA(4,5)=.997114),	LNG 112
4(ATA(4,6)=1.20955),(ATA(5,6)=1.13889)	LNG 113
DATA(ZATA(1,2)=1.01127),(ZATA(1,3)=.988608),(ZATA(1,4)=.983130),	LNG 114
1(ZATA(1,5)=.986978),(ZATA(1,6)=.953430),(ZATA(2,3)=.999961),	LNG 115
2(ZATA(2,4)=.972223),(ZATA(2,5)=.998886),(ZATA(2,6)=.939622),	LNG 116
3(ZATA(3,4)=.985547),(ZATA(3,5)=1.03099),(ZATA(3,6)=.912209),	LNG 117
4(ZATA(4,5)=.976416),(ZATA(4,6)=.849200),(ZATA(5,6)=.857310)	LNG 118
L=IBASE	LNG 119
N=8	LNG 120
N1=N-1	LNG 121
PC(7)=33.37	LNG 122
PC(8)=31.57	LNG 123
TC(7)=469.6	LNG 124
TC(8)=460.39	LNG 125
VC(7)=304.	LNG 126

VC(8)=306.	LNG 127
W(7)=72.15139	LNG 128
W(8)=72.15139	LNG 129
AC(7)=.234320	LNG 130
AC(8)=.288886	LNG 131
ATA(1,7)=ATA(1,8)=1.06	LNG 132
ATA(2,7)=ATA(2,8)=1.02	LNG 133
ATA(3,7)=ATA(3,8)=1.01	LNG 134
ATA(4,7)=ATA(4,8)=1.	LNG 135
ATA(5,7)=ATA(5,8)=1.	LNG 136
ATA(6,7)=ATA(6,8)=1.	LNG 137
ATA(7,8)=1.	LNG 138
ZATA(1,7)=ZATA(1,8)=.98	LNG 139
ZATA(2,7)=ZATA(2,8)=.99	LNG 140
ZATA(3,7)=ZATA(3,8)=.99	LNG 141
ZATA(4,7)=ZATA(4,8)=.99	LNG 142
ZATA(5,7)=ZATA(5,8)=.99	LNG 143
ZATA(6,7)=ZATA(6,8)=.99	LNG 144
ZATA(7,8)=1.	LNG 145
DO 3 J=1,N1	LNG 146
J1=J+1	LNG 147
DO 3 K=J1,N	LNG 148
ZATA(K,J)=ZATA(J,K)	LNG 149
3 ATA(K,J)=ATA(J,K)	LNG 150
DO 4 I=1,N	LNG 151
4 ATA(I,I)=ZATA(I,I)=1.	LNG 152
PCO=PC(L)	LNG 153
VCO=VC(L)	LNG 154
TCO=TC(L)	LNG 155
OMEGO=AC(L)	LNG 156
ZCO=PCO*VCO/TCO/R	LNG 157
DO 5 I=1,N	LNG 158
5 ZC(I)=PC(I)*VC(I)/TC(I)/R	LNG 159
RETURN	LNG 160
END	LNG 161
SUBROUTINE DATA CH4	LNG 162
C INITIALIZES THE EQUATION OF STATE CONSTANTS TO METHANE	LNG 163
DIMENSION G(32),VP(9),GI(11)	LNG 164
COMMON/DATA/G,R,GAMMA,VP,DTP	LNG 165
DIMENSION A(10)	LNG 166
COMMON/SATC/A	LNG 167
R=.08205616	LNG 168
GAMMA=-.0096	LNG 169
A(1)=190.555	LNG 170
A(2)=10.23	LNG 171
A(3)=18.404156472	LNG 172
A(4)=7.3498921512	LNG 173
A(5)=-1.4313160833	LNG 174
A(6)=A(7)=0.0	LNG 175
G( 1)=-.187027997685E-01	LNG 176
G( 2)=.103387108009E+01	LNG 177
G( 3)=-.155387625619E+02	LNG 178
G( 4)=.772311478564E+03	LNG 179
G( 5)=-.377103300895E+05	LNG 180
G( 6)=.846818843475E-03	LNG 181
G( 7)=-.496415884529E+00	LNG 182
G( 8)=.869909352414E+02	LNG 183
G( 9)=-.322821592493E+05	LNG 184
G(10)=-.395843026318E-04	LNG 185
G(11)=.266772318035E-01	LNG 186
G(12)=-.304010057839E+01	LNG 187
G(13)=.191584507536E-03	LNG 188
G(14)=-.195587933458E-03	LNG 189

G(15)= .607479967879E+01	LNG 190
G(16)= -.529609525984E-03	LNG 191
G(17)= .152264286004E-04	LNG 192
G(18)= -.109952182842E-01	LNG 193
G(19)= .191395549929E-03	LNG 194
G(20)= .386470003746E+05	LNG 195
G(21)= -.157930582612E+07	LNG 196
G(22)= .195270144401E+03	LNG 197
G(23)= .165996081629E+07	LNG 198
G(24)= .603051146711E+00	LNG 199
G(25)= .376485162808E+02	LNG 200
G(26)= .125593680622E-02	LNG 201
G(27)= -.343570032513E+02	LNG 202
G(28)= -.540945094139E-05	LNG 203
G(29)= .185622284663E-02	LNG 204
G(30)= .770786979245E-08	LNG 205
G(31)= -.286868318650E-05	LNG 206
G(32)= .372376961647E-04	LNG 207
VP(1)=4.77748580	LNG 208
VP(2)=1.76065363	LNG 209
VP(3)= -.56788894	LNG 210
VP(4)=1.32786231	LNG 211
VP(5)=1.5	LNG 212
VP(6)=.1159	LNG 213
VP(7)=90.68	LNG 214
VP(8)=190.555	LNG 215
END	LNG 216
FUNCTION FIND M(P,T,D)	LNG 217
C SOLVES THE EQUATION OF STATE OF METHANE FOR DENSITY GIVEN P AND T	LNG 218
DD=D	LNG 219
TT=T	LNG 220
DO 10 I=1,50	LNG 221
CALL PRESS(PP,DD,TT)	LNG 222
P2=PP	LNG 223
IF (ABS (P-P2)-1.E-7*P)20,20,1	LNG 224
1 CALL DPDD(PP,DD,TT)	LNG 225
DP=PP	LNG 226
CORR=(P2-P)/DP	LNG 227
IF (ABS (CORR)-1.E-7*DD)20,20,10	LNG 228
10 DD=DD-CORR	LNG 229
FIND M=0	LNG 230
RETURN	LNG 231
20 FIND M=DD	LNG 232
RETURN	LNG 233
END	LNG 234
FUNCTION SATL(T)	LNG 235
C CALCULATES THE SATURATED LIQUID DENSITY OF METHANE	LNG 236
DIMENSION A(10)	LNG 237
COMMON/SATC/A	LNG 238
IF(T.GT.A(1))GO TO 1	LNG 239
X=(1.-T/A(1))	LNG 240
SATL=A(2)+A(3)*X**(.35)+A(4)*X+A(5)*X**(4./3.)+A(6)*X**(5./3.	LNG 241
1)+A(7)*X**2	LNG 242
SATL=SATL/1000.	LNG 243
RETURN	LNG 244
1 SATL=1.E20	LNG 245
RETURN	LNG 246
END	LNG 247
SUBROUTINE PROPS(PP,DD,TT)	LNG 248
C EQUATION OF STATE FOR METHANE AND NITROGEN	LNG 249
DIMENSION X(33)	LNG 250
DIMENSION B(33),G(32)	LNG 251
EQUIVALENCE (B,X)	LNG 252

COMMON/DATA/G,R,GAMMA	LNG 253
DATA(ID=1)	LNG 254
DATA(IZ=1)	LNG 255
1 CONTINUE	LNG 256
D=DD	LNG 257
P=PP	LNG 258
T=TT	LNG 259
GM=GAMMA	LNG 260
D2=D*D	LNG 261
D3=D2*D	LNG 262
D4=D3*D	LNG 263
D5=D4*D	LNG 264
D6=D5*D	LNG 265
D7=D6*D	LNG 266
D8=D7*D	LNG 267
D9=D8*D	LNG 268
D10=D9*D	LNG 269
D11=D10*D	LNG 270
D12=D11*D	LNG 271
D13=D12*D	LNG 272
TS=SQRT (T)	LNG 273
T2=T*T	LNG 274
T3=T2*T	LNG 275
T4=T3*T	LNG 276
T5=T4*T	LNG 277
F=EXP (GM*D2)	LNG 278
GO TO (100,200),K	LNG 279
ENTRY PRESS	LNG 280
K=1	LNG 281
GO TO 1	LNG 282
100 CONTINUE	LNG 283
B( 1)=D2*T	LNG 284
B( 2)=D2*TS	LNG 285
B( 3)=D2	LNG 286
B( 4)=D2/T	LNG 287
B( 5)=D2/T2	LNG 288
B( 6)=D3*T	LNG 289
B( 7)=D3	LNG 290
B( 8)=D3/T	LNG 291
B( 9)=D3/T2	LNG 292
B(10)=D4*T	LNG 293
B(11)=D4	LNG 294
B(12)=D4/T	LNG 295
B(13)=D5	LNG 296
B(14)=D6/T	LNG 297
B(15)=D6/T2	LNG 298
B(16)=D7/T	LNG 299
B(17)=D8/T	LNG 300
B(18)=D8/T2	LNG 301
B(19)=D9/T2	LNG 302
B(20)=D3*F/T2	LNG 303
B(21)=D3*F/T3	LNG 304
B(22)=D5*F/T2	LNG 305
B(23)=D5*F/T4	LNG 306
B(24)=D7*F/T2	LNG 307
B(25)=D7*F/T3	LNG 308
B(26)=D9*F/T2	LNG 309
B(27)=D9*F/T4	LNG 310
B(28)=D11*F/T2	LNG 311
B(29)=D11*F/T3	LNG 312
B(30)=D13*F/T2	LNG 313
B(31)=D13*F/T3	LNG 314
B(32)=D13*F/T4	LNG 315



102	P=0	LNG	316
	DO 101 I=1,32	LNG	317
101	P=P+B(I)*G(I)	LNG	318
	P=P+R*D*T	LNG	319
	PP=P	LNG	320
	RETURN	LNG	321
	ENTRY DPDD	LNG	322
	K=2	LNG	323
	GO TO 1	LNG	324
200	CONTINUE	LNG	325
	F1=2.00*F*GM*D	LNG	326
	F21=3.000*F*D2 +F1*D3	LNG	327
	F22=5.000*F*D4 +F1*D5	LNG	328
	F23=7.000*F*D6 +F1*D7	LNG	329
	F24=9.000*F*D8 +F1*D9	LNG	330
	F25=11.00*F*D10+F1*D11	LNG	331
	F26=13.00*F*D12+F1*D13	LNG	332
	B( 1)=2.00*D*T	LNG	333
	B( 2)=2.00*D*TS	LNG	334
	B( 3)=2.00*D	LNG	335
	B( 4)=2.00*D/T	LNG	336
	B( 5)=2.00*D/T2	LNG	337
	B( 6)=3.00*D2*T	LNG	338
	B( 7)=3.00*D2	LNG	339
	B( 8)=3.00*D2/T	LNG	340
	B( 9)=3.00*D2/T2	LNG	341
	B(10)=4.00*D3*T	LNG	342
	B(11)=4.00*D3	LNG	343
	B(12)=4.00*D3/T	LNG	344
	B(13)=5.00*D4	LNG	345
	B(14)=6.00*D5/T	LNG	346
	B(15)=6.00*D5/T2	LNG	347
	B(16)=7.00*D6/T	LNG	348
	B(17)=8.00*D7/T	LNG	349
	B(18)=8.00*D7/T2	LNG	350
	B(19)=9.00*D8/T2	LNG	351
	B(20)=F21/T2	LNG	352
	B(21)=F21/T3	LNG	353
	B(22)=F22/T2	LNG	354
	B(23)=F22/T4	LNG	355
	B(24)=F23/T2	LNG	356
	B(25)=F23/T3	LNG	357
	B(26)=F24/T2	LNG	358
	B(27)=F24/T4	LNG	359
	B(28)=F25/T2	LNG	360
	B(29)=F25/T3	LNG	361
	B(30)=F26/T2	LNG	362
	B(31)=F26/T3	LNG	363
	B(32)=F26/T4	LNG	364
202	P=0	LNG	365
	DO 201 I=1,32	LNG	366
201	P=P+B(I)*G(I)	LNG	367
	P=P+R*T	LNG	368
	PP=P	LNG	369
	RETURN	LNG	370
	END	LNG	371
	SUBROUTINE DATA N2	LNG	372
C	INITIALIZES THE EQUATION OF STATE CONSTANTS TO NITROGEN	LNG	373
	DIMENSION G(32),VP(9),GI(11)	LNG	374
	COMMON/DATA/G,R,GAMMA,VP,DTP	LNG	375
	R=8.20539E-2	LNG	376
	GAMMA=-.0056	LNG	377
	G( 1)= 0.136224769272827E-02	LNG	378



G( 2)=	0.107032469908591E 00	LNG 379
G( 3)=	-0.243900721871413E 01	LNG 380
G( 4)=	0.341007449376470E 02	LNG 381
G( 5)=	-0.422374309466167E 04	LNG 382
G( 6)=	0.105098600246494E-03	LNG 383
G( 7)=	-0.112594826522081E-01	LNG 384
G( 8)=	0.142600789270907E-03	LNG 385
G( 9)=	0.184698501609007E 05	LNG 386
G(10)=	0.811140082588776E-07	LNG 387
G(11)=	0.233011645038006E-02	LNG 388
G(12)=	-0.507752586350986E 00	LNG 389
G(13)=	0.485027881931214E-04	LNG 390
G(14)=	-0.113656764115364E-02	LNG 391
G(15)=	-0.707430273540575E 00	LNG 392
G(16)=	0.751706648852680E-04	LNG 393
G(17)=	-0.111614119537424E-05	LNG 394
G(18)=	0.368796562233495E-03	LNG 395
G(19)=	-0.201317691347729E-05	LNG 396
G(20)=	-0.169717444755949E 05	LNG 397
G(21)=	-0.119719240044192E 06	LNG 398
G(22)=	-0.975218272038281E 02	LNG 399
G(23)=	0.554639713151823E 05	LNG 400
G(24)=	-0.179920450443470E 00	LNG 401
G(25)=	-0.256582926077184E 01	LNG 402
G(26)=	-0.413707715090789E-03	LNG 403
G(27)=	-0.256245415300293E 00	LNG 404
G(28)=	-0.124222373740063E-06	LNG 405
G(29)=	0.103556535840165E-04	LNG 406
G(30)=	-0.538699166558303E-09	LNG 407
G(31)=	-0.757415412839596E-08	LNG 408
G(32)=	0.585367172069521E-07	LNG 409
VP(1)=5.1113192094 \$ VP(2)=6.482667539E-1		LNG 410
VP(3)=-1.5108730916E-1 \$ VP(4)=7.4028493342E-1		LNG 411
VP(5)=1.5 \$ VP(6)=-.123 \$ VP(7)=63.15 \$ VP(8)=126.26		LNG 412
VP(9)=0.0		LNG 413
DTP=31.0		LNG 414
RETURN \$ END		LNG 415
FUNCTION VPN(TT)		LNG 416
C CALCULATES THE VAPOR PRESSURE OF BOTH METHANE AND NITROGEN		LNG 417
DIMENSION G(32),VP(9)		LNG 418
COMMON/DATA/G,R,GAMMA,VP,DTP		LNG 419
T=TT		LNG 420
X=(1.-VP(7)/T)/((1.-VP(7)/VP(8)))		LNG 421
VPN=VP(6)*EXP(VP(1)*X+VP(2)*X*X+VP(3)*X**3+VP(9)*X**4+VP(4)*X*		LNG 422
1(1.-X)**VP(5))		LNG 423
RETURN		LNG 424
END		LNG 425
FUNCTION RODEN(P,T,X)		LNG 426
C THE HARD SPHERE MODEL,SEE SUBROUTINE FM FOR THE ARGUMENT LIST		LNG 427
DIMENSION X(10)		LNG 428
CALL FM(P,T,X,V,G)		LNG 429
D=28.		LNG 430
V=V+V IDEL(P,D,T,X)		LNG 431
RODEN=1000./V		LNG 432
RETURN		LNG 433
END		LNG 434
SUBROUTINE FM(Q,T,X,V9,G9)		LNG 435
C PREDICTION OF EXCESS PROPERTIES WITH LHW POTENTIAL, COMP 1-METH		LNG 436
C ANE,COMP 2-ETHANE,COMP 3-PROPANE,COMP 4 N-BUTANE,COMP 5-I-BUT		LNG 437
C ANE, COMP 6-NITROGEN, COMP 7-NORMAL PENTANE,COMP 8-ISOPENTANE		LNG 438
C ADV CRYO ENGR. VOL. 19 (1973)-REPROGRAMED BY R. MCCARTY,2/22/74		LNG 439
C ARGUMENTS ARE X-MOLE FRACTIONS,T-TEMPERATURE,Q-PRESSURE,V9-EXCESS		LNG 440
C VOLUME,G9-EXCESS GIBBS ENERGY, FIRST THREE ARE INPUT, LAST TWO ARE		LNG 441

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C      OUTPUT,INPUT IS IN KELVINS AND BAR,                                LNG 442
      DIMENSION A(8),B(8),S(8),V(8),G(8),C(8,8),D(8,8),K(8,8),          LNG 443
      1J(8,8),E(8),Y(8),O(8),X(10)                                       LNG 444
      TYPE REAL J,K,N1                                                    LNG 445
      DATA(KEY=1)                                                         LNG 446
      DATA(J(1,2)=-.00388616),(J(1,3)=-.0120932),(J(1,4)=-.0231577),   LNG 447
      A(J(1,5)=-.0238349),(J(1,6)=-.00997547),(J(1,7)=-.0326),(J(1,8)=   LNG 448
      B-.0458),(J(2,4)=-.00400910),(J(2,5)=-.00812712),(J(2,6)=-.0143976) LNG 449
      C,(J(2,7)=-.003),(J(2,8)=-.004),(J(3,4)=+.0007615710),(J(3,5)=-.003 LNG 450
      D83743),(J(3,6)=-.024014),(J(3,7)=0.0),(J(3,8)=0.0),(J(4,5)=.002221 LNG 451
      E50),(J(4,7)=.0),(J(4,8)=.0),(J(5,6)=-.0576043),(J(4,6)=-.0576043), LNG 452
      F(J(5,7)=.0),(J(5,8)=0.0),(J(6,7)=-.04),(J(6,8)=-.05),(J(7,8)=.0)   LNG 453
      G,(J(2,3)=-.002162)                                                LNG 454
      DATA(K(1,2)=.00298830),(K(1,3)=.0597378),(K(1,4)=.110893),(      LNG 455
      AK(1,5)=.100298),(K(1,6)=.0197290),(K(1,7)=.14),(K(1,8)=.1745),     LNG 456
      B(K(2,4)=.0677703),(K(2,5)=.0346632),(K(2,6)=.0529034),(K(2,7)=   LNG 457
      C.02),(K(2,8)=.03),(K(3,4)=.0249291),(K(3,5)=-.00838212),(K(4,5)=.0 LNG 458
      D199213),(K(4,6)=.154365),(K(4,7)=.0),(K(4,8)=.0),(K(5,6)=.154365) LNG 459
      E,(K(5,7)=.0),(K(5,8)=.0),(K(6,7)=.15),(K(6,8)=.18),(K(7,8)=0.0)   LNG 460
      F,(K(2,3)=.014527),(K(3,6)=.14719),(K(3,7)=0.0),(K(3,8)=0.0)      LNG 461
      DATA(S=3.676E-8,4.158E-8,4.644E-8,5.051E-8,5.056E-8,3.546E-8,    LNG 462
      15.389E-8,5.706E-8)                                                LNG 463
      DATA(A=2.755E+5,7.773E+5,14.165E+5,22.733E+5,21.279E+5,1.718E+5,  LNG 464
      130.550E+5,42.946E+5)                                             LNG 465
      DATA(O=1.,1.5,1.67,1.83,1.79,1.03,1.91,2.11)                   LNG 466
C      THESE ARE THE ACENTRICITY FACTORS (FOR MOLECULAR SHAPES,ETC.) *** LNG 467
      DATA(Y=35.,45.,60.,75.,75.,40.,90.,105.)                       LNG 468
      DATA(P1=3.14159),(N1=6.025E+23),(R=8.3143)                     LNG 469
      IF(KEY.EQ.0)GO TO 1                                               LNG 470
      KEY=0                                                              LNG 471
      DO 2 I=1,8                                                         LNG 472
      J(I,I)=0.0                                                         LNG 473
      K(I,I)=0.0                                                         LNG 474
      DO 2 M=I,8                                                         LNG 475
      J(M,I)=J(I,M)                                                     LNG 476
      2 K(M,I)=K(I,M)                                                   LNG 477
      1 CONTINUE                                                         LNG 478
      P=Q*.1                                                            LNG 479
      IW=8                                                              LNG 480
      DO 10 I=1,IW                                                       LNG 481
      10 E(I)=O(I)                                                       LNG 482
      DO 15 I=1,IW                                                       LNG 483
      15 B(I)=(2./3.)*P1*N1*S(I)**3                                     LNG 484
      DO 20 I=1,IW                                                       LNG 485
      DO 20 M=1,IW                                                       LNG 486
      D(I,M) = (((B(I)**( 1./3. ) + B(M) ** ( 1./3.))/ 2. )*(1.-J(I,M) )) LNG 487
      1 **3                                                             LNG 488
      20 C( I , M ) = ( 1. - K( I , M ) ) * ( A(I)* A(M) ) ** ( 1. / 2. ) * LNG 489
      1( D( I , M ) **2 / ( B(I) * B(M) ) ) ** ( 1./2. )              LNG 490
      A2=0 $ B2=0                                                       LNG 491
      E2=0 $ V2=0                                                       LNG 492
      DO 25 I=1,IW                                                       LNG 493
      V2=V2+X(I)*Y(I)                                                   LNG 494
      DO 25 M=1,IW                                                       LNG 495
      E2=E2+X(I)*X(M)*(E(I)+E(M))/2.                                    LNG 496
      A2 = A2 + X(I) * X(M) * C( I , M )                               LNG 497
      25 B2=B2+X(I)*X(M)*D(I,M)                                         LNG 498
      V6=V2                                                             LNG 499
      DO 30 I = 1, IW                                                    LNG 500
      A3 = A(I)                                                         LNG 501
      B3 = B(I)                                                         LNG 502
      E3 = E(I)                                                         LNG 503
      V2 = Y(I)                                                         LNG 504

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V1 = FIND V1( A3, B3, E3, R, V2, P, T) LNG 505
G1 = FIND G1( A3, B3, E3, R, V2, P, T) LNG 506
V(I) = V1 LNG 507
30 G(I) = G1 LNG 508
A3=A2 LNG 509
B3=B2 LNG 510
E3=E2 LNG 511
V2=V6 LNG 512
V1=FIND V1(A3,B3,E3,R,V2,P,T) LNG 513
C G1=FIND G1(A3,B3,E3,R,V2,P,T) LNG 514
V7=V1 LNG 515
G7=G1 LNG 516
C EXCEAS VOLUME AND GIBS ENERGY LNG 517
V9=0 LNG 518
G9=0 LNG 519
DO 35 I=1,IW LNG 520
V9=V9-X(I)*V(I) LNG 521
35 G9=G9-X(I)*G(I) LNG 522
W9 = -V9 LNG 523
H9 = -G9 LNG 524
V9=V9+V7 LNG 525
G9=G9+G7 LNG 526
VIWW=V(IWW) LNG 527
RETURN LNG 528
END LNG 529
FUNCTION FIND V1( A3, B3, E3,R, V2, P, T ) LNG 530
C SOLVES THE HARD SPHERE EQUATION OF STATE FOR VOLUME GIVEN P AND T LNG 531
C A2 IS THE CONSTANT A, B2 IS THE CONSTANT B, E3 IS THE ACENTRICITY LNG 532
INDEX = 0 LNG 533
1 V1 = V2 LNG 534
X1 = B3 / ( 4. * V1 ) LNG 535
F2 = (( 1. + X1 + X1**2) / ( 1. - X1)**3) * E3 -A3 / ( V1 * R * T ) LNG 536
1- (P * V1) / ( R*T) LNG 537
F3 = A3 / (R*T*V1**2) - P/ (R*T) LNG 538
F3 = F3 - ((( X1 + 2.* X1**2) * ( 1 - X1) + 3* ( 1. + X1 + X1**2) LNG 539
1* X1) / (( 1. - X1 )**4 * V1)) * E3 LNG 540
V2 = V1 - F2/ F3 LNG 541
IF( ABS( (V2 - V1) / V2) .LT. .00001 ) GO TO 2 LNG 542
INDEX = INDEX + 1 LNG 543
IF( INDEX .LT. 250 ) GO TO 1 LNG 544
2 V1 = V2 LNG 545
FIND V1 = V2 LNG 546
RETURN LNG 547
END LNG 548
FUNCTION FIND G1( A3, B3, E3,R, V2, P, T ) LNG 549
C CALCULATES THE GIBBS FREE ENERGY FOR THE HARD SPHERE EOS LNG 550
V1 = V2 LNG 551
X1 = B3 / ( 4. * V1 ) LNG 552
G1 =ALOG( 1. / ( 1. - X1 )) +(3. *X1) / ( 1. - X1) + ( 3. * X1**2) LNG 553
1 / (2. * ( 1. - X1 )**2) LNG 554
G1 = G1 - A3 / ( E3 * R * T*V1 ) + ( P * V1) / ( E3 * R * T ) -1.0 LNG 555
1 - ALOG( V1 ) LNG 556
G1 = R * T * E3 * G1 LNG 557
FIND G1 = G1 LNG 558
RETURN LNG 559
END LNG 560
FUNCTION EXCESS(P,DD,T,X) LNG 561
C CALCULATES THE EXCESS OR IDEAL VOLUM DEPENDING ON THE ENTRY LNG 562
DIMENSION X(10),F(10) LNG 563
KR=0 LNG 564
GO TO 1 LNG 565
ENTRY V IDEL LNG 566
KR=1 LNG 567

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1	CONTINUE	LNG 568
	CALL ZERO(F)	LNG 569
	IF(X(1).LE..000001)GO TO 2	LNG 570
	CALL DATA CH4	LNG 571
	IF(T.GT.190.555)GO TO 12	LNG 572
	PM=VPN(T)+.00001	LNG 573
	DELP=P-PM	LNG 574
	D=SAT(T,1)	LNG 575
	CALL DPDD(DP,D,T)	LNG 576
	DELD=DELP/DP	LNG 577
	D=DELD+D	LNG 578
	F(1)=X(1)*1000./D	LNG 579
	GO TO 2	LNG 580
12	D=FINDD M(P,T,DD)	LNG 581
	IF(D.LE.0.0)D=1000.	LNG 582
	F(1)=X(1)*1000./D	LNG 583
2	IF(X(2).LE..000001)GO TO 3	LNG 584
	F(2)=X(2)*1000./SAT(T,2)	LNG 585
3	IF(X(3).LE..000001)GO TO 4	LNG 586
	F(3)=X(3)*1000./SAT(T,3)	LNG 587
4	IF(X(4).LE..00001)GO TO 5	LNG 588
	F(4)=X(4)*1000./SAT(T,4)	LNG 589
5	IF(X(5).LE..00001)GO TO 6	LNG 590
	F(5)=X(5)*1000./SAT(T,5)	LNG 591
6	IF(X(7).LE..00001)GO TO 61	LNG 592
	F(7)=X(7)*1000./SAT(T,7)	LNG 593
61	IF(X(8).LE..00001)GO TO 62	LNG 594
	F(8)=X(8)*1000./SAT(T,8)	LNG 595
62	IF(X(6).LE..00001)GO TO 8	LNG 596
	CALL DATA N2	LNG 597
	IF(T.GT.126.6)GO TO 7	LNG 598
	PN=VPN(T)+.000001	LNG 599
	DELP=P-PN	LNG 600
	D=SAT(T,6)	LNG 601
	CALL DPDD(DP,D,T)	LNG 602
	F(6)=X(6)*1000./(D+DELP/DP)	LNG 603
	GO TO 8	LNG 604
7	D=FINDD M(P,T,DD)	LNG 605
	IF(D.LE.0.0)D=1000.	LNG 606
	F(6)=X(6)*1000./D	LNG 607
8	V=1000./DD	LNG 608
	VS=0	LNG 609
	DO 21 I=1,8	LNG 610
21	VS=VS+F(I)	LNG 611
	EXCESS=V-VS	LNG 612
	IF(KR.GT.0)EXCESS=VS	LNG 613
	CALL DATA CH4	LNG 614
	RETURN	LNG 615
	END	LNG 616
	SUBROUTINE ZERO(X)	LNG 617
C	INITIALIZES THE COMPONENT MATRIX TO 0	LNG 618
	DIMENSION X(10)	LNG 619
	DO 1 I=1,10	LNG 620
1	X(I)=0.0	LNG 621
	RETURN	LNG 622
	END	LNG 623
	FUNCTION FMKM(T,X)	LNG 624
C	THE REVISED KLOSEK AND MCKINLEY METHOD,THE INPUT IS TEMPERATURE	LNG 625
C	AND THE COMPONENT MATRIX. TEMPERATURE IS IN KELVIN,OUTPUT IS	LNG 626
C	DENSITY IN MOLES PER LITER. THE ALLOWABLE COMPONENTS ARE C1,C2,C3	LNG 627
C	NC4,IC4,N2,NC5,IC5 IN THAT ORDER. THIS METHOD SHOULD NOT BE USED	LNG 628
C	FOR MIXTURES WITH LESS THAN 60% METHANE, OR FOR MIXTURES CONTAININ	LNG 629
C	MORE THAN 4% NITROGEN OR MORE THAN 4% EACH OF NC4 OR IC4 OR MORE	LNG 630



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C      THAN 2% TOTAL OF NC5 AND IC5.                                LNG 631
      DIMENSION TM(100),TN(100),X(10),Q(8)                        LNG 632
      DATA(Q=16.04303,30.07012,44.09721,58.1243,58.1243,28.0134,72.1513 LNG 633
19,72.15139)                                                         LNG 634
      DATA((TM(I),I=1,10)=-.005,.12,.22,.34,.43,.515,.595,.66,.725,.795) LNG 635
      DATA((TM(I),I=11,20)=-.006,.135,.26,.38,.5,.59,.665,.74,.81,.885) LNG 636
      DATA((TM(I),I=21,30)=-.007,.15,.3,.425,.575,.675,.755,.83,.91,.99) LNG 637
      DATA((TM(I),I=31,40)=-.007,.165,.34,.475,.635,.735,.84,.92,1.045, LNG 638
A1.12)                                                                LNG 639
      DATA((TM(I),I=41,50)=-.008,.19,.375,.535,.725,.835,.95,1.055,1.155 LNG 640
A,1.245)                                                            LNG 641
      DATA((TM(I),I=51,60)=-.009,.22,.44,.61,.81,.945,1.065,1.18,1.28, LNG 642
A1.38)                                                            LNG 643
      DATA((TM(I),I=61,70)=-.01,.25,.5,.695,.92,1.055,1.205,1.33,1.45, LNG 644
A1.55)                                                            LNG 645
      DATA((TM(I),I=71,80)=-.013,.295,.59,.795,1.035,1.21,1.385,1.525, LNG 646
11.64,1.75)                                                       LNG 647
      DATA((TM(I),I=81,90)=-.015,.345,.7,.92,1.2,1.37,1.555,1.715, LNG 648
A1.86,1.99)                                                       LNG 649
      DATA((TM(I),I=91,100)=-.017,.4,.825,1.06,1.39,1.59,1.8,1.95, LNG 650
A2.105,2.272)                                                     LNG 651
      DATA((TN(I),I=1,10)=-.004,.1,.22,.35,.5,.6,.69,.78,.86,.95) LNG 652
      DATA((TN(I),I=11,20)=-.005,.12,.28,.43,.59,.71,.83,.94,1.05,1.14) LNG 653
      DATA((TN(I),I=21,30)=-.007,.16,.34,.49,.64,.79,.94,1.08,1.17,1.27) LNG 654
      DATA((TN(I),I=31,40)=-.01,.24,.42,.61,.75,.91,1.05,1.19,1.33,1.45) LNG 655
      DATA((TN(I),I=41,50)=-.015,.32,.59,.77,.92,1.07,1.22,1.37,1.52, LNG 656
11.71)                                                            LNG 657
      DATA((TN(I),I=51,60)=-.024,.41,.72,.95,1.15,1.22,1.3,1.45,1.65,2.) LNG 658
      DATA((TN(I),I=61,70)=-.032,.6,.91,1.23,1.43,1.63,1.85,2.08,2.3, LNG 659
12.45)                                                            LNG 660
      DATA((TN(I),I=71,80)=-.043,.71,1.13,1.48,1.73,1.98,2.23,2.48,2.75 LNG 661
1,2.9)                                                            LNG 662
      DATA((TN(I),I=81,90)=-.058,.95,1.46,1.92,2.2,2.42,2.68,3., LNG 663
A3.32,3.52)                                                       LNG 664
      DATA((TN(I),I=91,100)=-.075,1.3,2.,2.4,2.6,3.,3.4,3.77, LNG 665
A3.99,4.23)                                                       LNG 666
      IF(X(1).LT..00001)GO TO 20                                    LNG 667
      AW=0.0                                                         LNG 668
      DO 1 I=1,8                                                     LNG 669
1  AW=AW+X(I)*Q(I)                                                  LNG 670
      VI=VIDEAL(T,X)                                                LNG 671
      J=1                                                            LNG 672
      IF(T.GE.95.)J=11                                              LNG 673
      IF(T.GE.100.)J=21                                             LNG 674
      IF(T.GE.105.)J=31                                             LNG 675
      IF(T.GE.110.)J=41                                             LNG 676
      IF(T.GE.115.)J=51                                             LNG 677
      IF(T.GE.120.)J=61                                             LNG 678
      IF(T.GE.125.)J=71                                             LNG 679
      IF(T.GE.130.)J=81                                             LNG 680
      JJ=J+9                                                         LNG 681
      W=15.                                                         LNG 682
      DO 5 I=J,JJ                                                    LNG 683
      W=W+1.                                                         LNG 684
      IF(AW.GT.W)GO TO 5                                            LNG 685
      GO TO 6                                                        LNG 686
5  CONTINUE                                                         LNG 687
      I=JJ                                                           LNG 688
6  DIF1=AW-W                                                         LNG 689
      J=I-1                                                         LNG 690
      FK=(TM(I)-TM(J))*DIF1+TM(I)                                    LNG 691
      FK1=(TM(I+10)-TM(J+10))*DIF1+TM(I+10)                        LNG 692
      IT=(T+.00001)/5.                                             LNG 693

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DIF2=T-IT*5
IF(T.GE.135.)DIF2=T-130.
IF(T.LT.90.)DIF2=T-90.
FK=FK+(FK1-FK)*DIF2/5.
IF(X(6).LT..0001)GO TO 17
FKN=(TN(I)-TN(J))*DIF1+TN(I)
FK1=(TN(I+10)-TN(J+10))*DIF1+TN(I+10)
FKN=FKN+(FK1-FKN)*DIF2/5.
FK=FK+(FKN-FK)*X(6)/.0425
17 FK=FK/1000.
FMKM=1./(VI-FK*X(1))
RETURN
20 FMKM=0.0
RETURN
END
FUNCTION V IDEAL(T,X)
C CALCULATES THE IDEAL VOLUME OF A MIXTURE FOR THE K AND M METHOD
DIMENSION X(10)
V=0
J=0
IF(X(6).GT..0001.AND.T.GT.115.)J=1
DO 10 I=1,8
IF(X(I).LE..000001)GO TO 10
IF(J.GT.0.AND.I.EQ.6)GO TO 10
V=V+X(I)/SAT(T,I)
10 CONTINUE
IF(J.EQ.1)V=V+X(6)/SATN2(T)
VIDEAL=V
RETURN
END
FUNCTION SATN2(T)
C CALCULATES A PSEUDO SATURATED LIQUID DENSITY FOR N2 ABOVE 115 K
IF(T.LT.115.)GO TO 1
DELT=(T-115.)
SATN2=SAT(115.,6)+DELT*(SAT(115.05,6)-SAT(114.95,6))/.1
RETURN
1 SATN2=SAT(T,6)
RETURN
END
FUNCTION SAT(T,I)
C CALCUALTES THE PURE FLUID DENSITIES FOR THE K AND M METHOD
C UNITS ARE DEG K AND MOLES/LITER
DIMENSION A(7,8)
DATA((A(I),I=1,7)=190.555,10.16,18.65812322,6.712030737,
1-.9472019702,0.0,0.0)
DATA((A(I),I=8,14)=305.33,
1 6.86,12.55205121,13.43284373,-19.00461066,11.07715985,0.0)
DATA((A(I),I=15,21)=369.82,5.,8.684458671,18.04085714,-29.46261356
1,16.43559114,0.0)
DATA((A(I),I=22,28)=425.16,3.92,
1 7.286062567,11.96307859,-19.87591962,
211.60211932,0.0)
DATA((A(I),I=29,35)=408.13,3.8,
1 7.657535400,8.145251283,-13.10582462,
28.145894091,0.0)
DATA((A(I),I=36,42)=126.2,11.21,19.39216835,26.01408462,-39.497587
191.23.32977312,0.0)
DATA((A(I),I=43,49)=469.6,3.285,-.0362004993,59.00202990,
1-93.44193819,43.66780833,0.0)
DATA((A(I),I=50,56)=460.39,3.271,2.946310456,35.50770979,
1-57.41242993,28.15898339,0.0)
IF(T.GT.A(1,I))GO TO 1
X=(1.-T/A(1,I))

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LNG 694  
 LNG 695  
 LNG 696  
 LNG 697  
 LNG 698  
 LNG 699  
 LNG 700  
 LNG 701  
 LNG 702  
 LNG 703  
 LNG 704  
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 LNG 706  
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 LNG 709  
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 LNG 742  
 LNG 743  
 LNG 744  
 LNG 745  
 LNG 746  
 LNG 747  
 LNG 748  
 LNG 749  
 LNG 750  
 LNG 751  
 LNG 752  
 LNG 753  
 LNG 754  
 LNG 755  
 LNG 756

[illegible]

COMMON/UNITS/ITC,IPC,T,TMAX,DT,P,PMAX,DP,PS	LNG 820
SWITCH = 0	LNG 821
1 CALL ZERO1	LNG 822
C THE NEXT VALUES ARE SET TO PREVENT UNDER/OVER FLOWS	LNG 823
TMAX = 100.0	LNG 824
DT = 100.0	LNG 825
PMAX = 100.0	LNG 826
	LNG 827
DP = 100.0	LNG 828
PS = 100.0	LNG 829
CALL INPUT(Q)	LNG 830
CODE=0	LNG 831
ITC=0	LNG 832
IPC=0	LNG 833
P=PIN/1.01325	LNG 834
T=TIN	LNG 835
IF(T.GT.(160.)) GO TO 19	LNG 836
8 CALL PZERO(CTMX)	LNG 837
MWM = 0	LNG 838
DO 9 I = 1,NC	LNG 839
MWM = MWM + (X(I)*MW(I))	LNG 840
9 CONTINUE	LNG 841
10 DO 11 I = 1,NC	LNG 842
TBI(I) = T / TS(I)	LNG 843
CALL VOLUME(TBI(I),VBI(I),JPUR,NC)	LNG 844
TBIM(I) = T / TSIM(I)	LNG 845
CALL VOLUME(TBIM(I),VBIM(I),JMX,NC)	LNG 846
11 CONTINUE	LNG 847
VM = 0.0	LNG 848
DO 12 I = 1,NC	LNG 849
VM = VM + (X(I)*VBIM(I)*VSIM(I))	LNG 850
12 CONTINUE	LNG 851
TMP = ( T * 1.8) - (459.67)	LNG 852
DO 13 I = 1,NC	LNG 853
VHOLD(I) = VBI(I)	LNG 854
VPHOLD(I) = VBIM(I)	LNG 855
PP = (P*TS(I))/US(I)	LNG 856
VBIMP(I) = VBI(I)	LNG 857
CALL PRES(PP,VBIMP(I),TBI(I),NC)	LNG 858
VBI(I) = VBIMP(I)	LNG 859
PP = (P * TSIM(I)) / USIM(I)	LNG 860
VBIMP(I) = VBIM(I)	LNG 861
CALL PRES(PP,VBIMP(I),TBIM(I),NC)	LNG 862
VBIM(I) = VBIMP(I)	LNG 863
13 CONTINUE	LNG 864
VMP = 0.0	LNG 865
DO 14 I = 1,NC	LNG 866
VMP = VMP + (X(I)*VBIMP(I)*VSIM(I))	LNG 867
PD(I) = VBIMP(I) * VSIM(I)	LNG 868
14 CONTINUE	LNG 869
DENS=MWM/VMP	LNG 870
DOUT=DENS*1000./MWM	LNG 871
RETURN	LNG 872
19 DOUT=0.0	LNG 873
RETURN	LNG 874
END	LNG 875
SUBROUTINE ZERO1	LNG 876
COMMON /RUN/A(361)	LNG 877
COMMON /DAT/B(224)	LNG 878
DO 1 I = 1,361	LNG 879
1 A(I) = 0.0	LNG 880
DO 2 I = 1,224	LNG 881
2 B(I) = 0.0	LNG 882

RETURN	LNG 883
END	LNG 884
SUBROUTINE INPUT(Q)	LNG 885
DIMENSION NNO( 8),SIGM( 8),EPSI( 8),LAMB( 8),NAME(2, 8),SNO( 8),MO	LNG 886
1L( 8),CT( 8),TCH( 8),VCH( 8),ECH( 8),AR( 8),AZ( 8),Q(8)	LNG 887
TYPE INTEGER CNT1,CNT2,SWITCH	LNG 888
TYPE REAL MWM,INCR,KIJ,MW,MOL,LAM,LAMB	LNG 889
COMMON /RUN/ID(12),X(12),NAM(2,12),C12(12,12),R(12),Z(12),S(12),VS	LNG 890
1(12),TS(12),US(12),SG(12),EP(12),LAM(12),NC,MW(12),AIJ(12,12),PST(	LNG 891
212),TCT(12)	LNG 892
COMMON /DAT/TBI(12),VBI(12),TBIM(12),VBIM(12),C(12),SGIM(12),SIM(1	LNG 893
12),EPIM(12),PIM(12),SIN(12),VSIM(12),USIM(12),TSIM(12),DVBIM(12),P	LNG 894
2D(12),PV(12),RHO(12),VBIMP(12),TTM(12),MWM,CNT1,CNT2,TOLD(12),INCR	LNG 895
3(12),SWITCH,JPC,JPCS,JMIX,DENS,VEX,HEX,GEX,TMP,VMP,TP	LNG 896
COMMON/PAR/KIJ( 8, 8),AJI( 8, 8)	LNG 897
DATA SIGM/	LNG 898
*0.991000,1.029000,1.155000,1.278000,1.388752,1.392995,	LNG 899
*1.47162,1.47217/	LNG 900
DATA EPSI/	LNG 901
*0.640000,0.909000,1.69800,2.237000,2.705262,2.545907,	LNG 902
*3.76195,3.68336/	LNG 903
DATA LAMB/	LNG 904
*1.053604,0.986325,1.227117,1.408519,1.473346,1.461676,	LNG 905
*1.698790,1.695012/	LNG 906
DATA MOL/	LNG 907
*28.01600,16.04200,30.06800,44.09400,58.12000,58.12000,	LNG 908
*72.146000,72.146000/	LNG 909
DATA SNO/	LNG 910
*10.00000,10.00000,10.00000,10.00000,10.00000,10.00000,	LNG 911
*10.00000,10.00000/	LNG 912
DATA CT/	LNG 913
*126.0600,190.5600,305.4300,369.8200,425.1600,408.0300,	LNG 914
*469.65,460.39/	LNG 915
DATA TCH/	LNG 916
*112.7699,170.9645,256.8311,294.7255,340.7867,322.6109,	LNG 917
*411.0100,403.3200/	LNG 918
DATA VCH/	LNG 919
*26.01846,29.04010,40.88188,54.60301,68.72075,69.44272,	LNG 920
*84.90197,84.99830/	LNG 921
DATA ECH/	LNG 922
*1393.000,1977.000,3695.000,4867.000,5886.652,5539.895,	LNG 923
*8186.000,8015.000/	LNG 924
DATA AR/	LNG 925
*1.000000,1.000000,1.000000,1.000000,1.000000,1.000000,	LNG 926
*1.000000,1.000000/	LNG 927
DATA AZ/	LNG 928
*10.00000,10.00000,10.00000,10.00000,10.00000,10.00000,	LNG 929
*10.00000,10.00000/	LNG 930
DO 4 I=2,6	LNG 931
4 X(I)=Q(I-1)	LNG 932
X(1)=Q(6)	LNG 933
X(7)=Q(7)	LNG 934
X(8)=Q(8)	LNG 935
NC=0	LNG 936
DO 1 I=1,8	LNG 937
IF(X(I).LE.0.0)GO TO 1	LNG 938
NC=NC+1	LNG 939
X(NC)=X(I)	LNG 940
ID(NC)=I	LNG 941
1 CONTINUE	LNG 942
DO 2 I = 1,NC	LNG 943
J = ID(I)	LNG 944
R(I) = AR(J)	LNG 945



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Z(I) = AZ(J) LNG 946
S(I) = SNO(J) LNG 947
TS(I) = TCH(J) LNG 948
US(I) = ECH(J) LNG 949
SG(I) = SIGM(J) LNG 950
TCT(I) = CT(J) LNG 951
EP(I) = EPSI(J) LNG 952
LAM(I) = LAMB(J) LNG 953
VS(I) = VCH(J) LNG 954
MW(I) = MOL(J) LNG 955
2 CONTINUE LNG 956
DO 3 I = 1,NC LNG 957
DO 3 J = 1,NC LNG 958
M = ID(I) LNG 959
L = ID(J) LNG 960
C12(I,J) = KIJ(M,L) LNG 961
C12(J,I) = KIJ(L,M) LNG 962
AIJ(I,J) = AJI(M,L) LNG 963
AIJ(J,I) = AJI(L,M) LNG 964
3 CONTINUE LNG 965
RETURN LNG 966
END LNG 967
END LNG 968
SUBROUTINE PZERO(TRT) LNG 969
TYPE INTEGER CNT1,CNT2,SWITCH LNG 970
TYPE REAL MWM,INCR,KIJ,MW,MOL,LAM,LAMB LNG 971
COMMON /RUN/ID(12),X(12),NAM(2,12),C12(12,12),R(12),Z(12),S(12),VS LNG 972
1(12),TS(12),US(12),SG(12),EP(12),LAM(12),NC,MW(12),AIJ(12,12),PST( LNG 973
212),TCT(12) LNG 974
COMMON /DAT/TBI(12),VBI(12),TBIM(12),VBIM(12),C(12),SGIM(12),SIM(1 LNG 975
12),EPIM(12),PIM(12),SIN(12),VSIM(12),USIM(12),TSIM(12),DVBIM(12),P LNG 976
2D(12),PV(12),RHO(12),VBIMP(12),TTM(12),MWM,CNT1,CNT2,TOLD(12),INCR LNG 977
3(12),SWITCH,JPC,JPCS,JMIX,DENS,VEX,HEX,GEX,TMP,VMP,TP LNG 978
COMMON/PAR/KIJ( 8, 8),AJI( 8, 8) LNG 979
COMMON /UNITS/ITC,IPC,T,TMAX,DT,P,PMAX,DP,PS LNG 980
IF(NC.EQ.1) TRT = T/TCT(1) LNG 981
IF(NC.GT.1) CALL TCM(TRT) LNG 982
IF(NC.GT.1) TRT = T/TRT LNG 983
IF(NC.EQ.1) CALL PURE(TRT,TRT,R(1),RXX) LNG 984
DO 1 I = 1,NC LNG 985
TCXX = T/TCT(I) LNG 986
IF(NC.GT.1) CALL PURE(TCXX,TRT,R(I),RTR) LNG 987
JPC = 0 LNG 988
IF(TRT.GT.(.87)) JPC = 2 LNG 989
IF(TRT.GT.(1.0)) JPC = 1 LNG 990
IF(NC.EQ.1) RTR = RXX LNG 991
C THERE ARE SEVERAL OTHER FORMS OF THESE EQUATIONS LNG 992
C REFER TO ORIGINAL ARTICLES FOR VALUES TO USE LNG 993
S(I) = RTR * Z(I) - 2. * RTR + 2. LNG 994
SIM(I) = S(I) LNG 995
SG(I) = (VS(I)/RTR ) ** (1./3.) LNG 996
EP(I) = US(I) / S(I) LNG 997
C(I) = LAM(I) * (2176./((185.6)*(1.98726))) LNG 998
TTM(I) = TS(I) LNG 999
TOLD(I) = TS(I) LNG1000
INCR(I) = (20.0) LNG1001
1 CONTINUE LNG1002
JPUR = 0 LNG1003
CNT1 = 1 LNG1004
CNT2 = 1 LNG1005
JMIX = 0 LNG1006
2 DEN = 0 LNG1007
LNG1008

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DO 3 I = 1,NC	LNG1009
DEN = DEN + (X(I) * SIM(I))	LNG1010
3 CONTINUE	LNG1011
DO 4 I = 1,NC	LNG1012
PIM(I) = (X(I) * SIM(I)) / DEN	LNG1013
4 CONTINUE	LNG1014
DO 6 I = 1,NC	LNG1015
SGIM(I) = 0.0	LNG1016
DO 6 J = 1,NC	LNG1017
M = I	LNG1018
N = J	LNG1019
IF(I.GT.J) M = J	LNG1020
IF(I.GT.J) N = I	LNG1021
IF(I.EQ.J) GO TO 5	LNG1022
TRUDX = (AIJ(N,M)/(T/TTM(I)))	LNG1023
IF(TRUDX.LT.(-180.0)) GO TO 5	LNG1024
SGIM(I) = SGIM(I) +((PIM(J) *(((SG(I)**(1./3.) + SG(J)**(1./3.))	LNG1025
1/((2.0) **3))* (AIJ(M,N) * EXP(AIJ(N,M)/(T/TTM(I))))	LNG1026
GO TO 6	LNG1027
5 SGIM(I) = SGIM(I) +(PIM(J) *(((SG(I)**(1./3.) + SG(J)**(1./3.))	LNG1028
1/((2.0) **3 )	LNG1029
6 CONTINUE	LNG1030
DEN = 0.0	LNG1031
DO 7 I = 1,NC	LNG1032
DEN = DEN + (PIM(I)* SGIM(I)* SGIM(I))	LNG1033
7 CONTINUE	LNG1034
DO 8 I = 1,NC	LNG1035
SIN(I) =(S(I) * (SGIM(I)**(2.))) / DEN	LNG1036
8 CONTINUE	LNG1037
	LNG1038
DO 9 I = 1,NC	LNG1039
TEST = (1.0) - (SIM(I) / SIN(I))	LNG1040
TEST = ABS (TEST)	LNG1041
IF(TEST.GT.(0.00001)) GO TO 10	LNG1042
9 CONTINUE	LNG1043
GO TO 12	LNG1044
10 DO 11 I = 1,NC	LNG1045
SIM(I) =(SIM(I) + SIN(I)) /(2.0)	LNG1046
11 CONTINUE	LNG1047
CNT1 = CNT1 + 1	LNG1048
IF(CNT1.GT.250) GO TO 12	LNG1049
GO TO 2	LNG1050
12 DO 14 I = 1,NC	LNG1051
SIM(I) = SIN(I)	LNG1052
EPIM(I) = 0.0	LNG1053
DO 14 J = 1,NC	LNG1054
K = I	LNG1055
L = J	LNG1056
IF(J.LT.I) K = J	LNG1057
IF(J.LT.I) L = I	LNG1058
IF(I.EQ.J) GO TO 13	LNG1059
TRUDX = C12(L,K) / (T/TTM(I))	LNG1060
IF(TRUDX.LT.(-180.0)) GO TO 13	LNG1061
EPIM(I) = EPIM(I) +((PIM(J) * SQRT(EP(I)*EP(J))))*	LNG1062
1(C12(K,L) * EXP((C12(L,K) /(T/TTM(I))))))	LNG1063
GO TO 14	LNG1064
13 EPIM(I) = EPIM(I) + (PIM(J) * SQRT(EP(I) * EP(J)))	LNG1065
14 CONTINUE	LNG1066
C ASSUME NUMBER OF MOLS OF MIXTURE = 1.0	LNG1067
DO 15 I = 1,NC	LNG1068
VSIM(I) = R(I) * (SGIM(I)**(3.0))	LNG1069
USIM(I) = SIM(I) * EPIM(I)	LNG1070
TSIM(I) = USIM(I) / ((1.98726) * C(I))	LNG1071

15 CONTINUE	LNG1072
DO 16 I = 1,NC	LNG1073
TEST = (1.0) - (TSIM(I)/TTM(I))	LNG1074
TEST = ABS(TEST)	LNG1075
IF(TEST.GT.(0.00001)) GO TO 17	LNG1076
16 CONTINUE	LNG1077
RETURN	LNG1078
17 DO 20 I = 1,NC	LNG1079
IF(TSIM(I) - TTM(I))18,20,19	LNG1080
18 TNEW = TTM(I) - INCR(I)	LNG1081
IF(TNEW.EQ.TOLD(I)) INCR(I) = INCR(I)/ (2.0)	LNG1082
IF(TNEW.EQ.TOLD(I)) GO TO 18	LNG1083
TOLD(I) = TTM(I)	LNG1084
TTM(I) = TNEW	LNG1085
GO TO 20	LNG1086
19 TNEW = TTM(I) + INCR(I)	LNG1087
IF(TNEW.EQ.TOLD(I)) INCR(I) = INCR(I) / (2.0)	LNG1088
IF(TNEW.EQ.TOLD(I)) GO TO 19	LNG1089
TOLD(I) = TTM(I)	LNG1090
TTM(I) = TNEW	LNG1091
20 CONTINUE	LNG1092
CNT2 = CNT2 + 1	LNG1093
IF(CNT2.GT.250) RETURN	LNG1094
GO TO 2	LNG1095
END	LNG1096
SUBROUTINE PURE(TZ,T,Z,R)	LNG1097
DIMENSION A(15)	LNG1098
IF(TZ.LT.(.8653525)) R = Z	LNG1099
IF(TZ.LT.(.8653525)) GO TO 2	LNG1100
IF(T.LT.(.8653525)) R = Z	LNG1101
IF(T.LT.(.8653525)) GO TO 2	LNG1102
IF(T.GT.(1.0)) TA = 1.0	LNG1103
IF(T.LE.(1.0)) TA = T	LNG1104
DATA A/.9184780,-.1530647,-.1090050,.8073883,1.441803,	LNG1105
1-10.85944,-6.041687,51.26758,.9062500,-108.9805,28.88672,	LNG1106
2106.1406,-45.78125,-38.43750,20.56250/	LNG1107
R = 0.0	LNG1108
TT = (TA - (.9267292)) / (.7267517E-01)	LNG1109
DO 1 K = 1,14	LNG1110
R = R + A(16-K)	LNG1111
R = R * TT	LNG1112
1 CONTINUE	LNG1113
R = R + A(1)	LNG1114
2 RETURN	LNG1115
END	LNG1116
SUBROUTINE VOLUME(T,V,J,NC)	LNG1117
TR = T	LNG1118
IF(TR.GT.(1.00)) GO TO 4	LNG1119
V = 0.5	LNG1120
	LNG1121
	LNG1122
1 VT = (1.0) + (0.1*TR*V**(4.0))	LNG1123
ERR = (1.0) - (V/VT)	LNG1124
TEST = ABS(ERR)	LNG1125
IF(TEST.LE.(0.00001)) GO TO 2	LNG1126
V = VT	LNG1127
GO TO 1	LNG1128
2 V = V**(3.0)	LNG1129
3 RETURN	LNG1130
4 V = ((((((10.06600 *TR)-24.79837)*TR)+23.260722)*TR)-6.686880)	LNG1131
GO TO 3	LNG1132
END	LNG1133
SUBROUTINE BETA(P,DV,T,V,KK)	LNG1134

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TYPE REAL AF,BF,CF,DF,EF,AG,BG,CG,DG,P,DV,T,V,F,F1,G,G1,DEN,EG    LNG1135
TYPE REAL C,B1,B2,B3                                                LNG1136
DIMENSION C(11)                                                       LNG1137
DATA AF/0.11566564E+02/,BF/-0.53510144E+01/,CF/-0.74598207E-01/,    LNG1138
1DF/0.67068653E+00/,EF/-0.11939887E+00/,AG/0.62721813E+01/,        LNG1139
2BG/0.47698365E+00/,CG/-0.16023080E+01/,DG/0.49837746E+00/,        LNG1140
3EG/-0.42639183E-01/                                                LNG1141
F = AF +(BF*V)+(CF*V*V)+(DF*V*V*V)+(EF*V*V*V*V)                    LNG1142
IF(F.LT.(0.0)) GO TO 1                                              LNG1143
F1 = (BF+(CF*V*2.E+00)+(DF*V*V*3.E+00)+(EF*V*V*V*4.E+00))*        LNG1144
1T * EXP(F)                                                           LNG1145
G = AG+(BG*V)+(CG*V*V)+(DG*V*V*V)+(EG*V*V*V*V)                    LNG1146
IF(G.LT.(0.00)) GO TO 1                                              LNG1147
G1 = (BG+(CG*V*2.E+00)+(DG*V*V*3.E+00+(EG*V*V*V*4.E+00))) * EXP(G    LNG1148
1)                                                                      LNG1149
G1 = -G1                                                              LNG1150
DEN = F1 + G1                                                         LNG1151
DV = (1.E+00) / DEN                                                  LNG1152
IF(DV.GT.(0.0).OR.DV.LT.(-1.0E+00)) GO TO 1                        LNG1153
GO TO 2                                                                LNG1154
DATA C/- .118659822E+02, -.5509099E+00, .4172102E+01, -.8996686E+00,    LNG1155
1-.1500376E+01, -.1958324E+00, -.2788988E+01, .5195283E+00,          LNG1156
2.3734878E+01, .1361904E+00, .4671948E-01/                          LNG1157
1 B1 = C(1) + C(2)*T + C(3)*V                                        LNG1158
B2 = C(4)*((V+C(5))**(2 ))*EXP((C(6))*((T+C(7))**(2 )))            LNG1159
B3 = C(11) * (V+C(8))**(C(9))/(T**(C(10)))                          LNG1160
DV = B1 + B2 + B3                                                    LNG1161
DV = EXP(DV)                                                         LNG1162
DV = - DV                                                            LNG1163
2 RETURN                                                              LNG1164
END                                                                    LNG1165
SUBROUTINE PRES(A,B,C,NC)                                             LNG1166
TYPE REAL H,P,V,T,K1,K2,K3,K4,VT,PT                                  LNG1167
P = A                                                                  LNG1168
V = B                                                                  LNG1169
T = C                                                                  LNG1170
H = 1.E-02                                                            LNG1171
IF(H.GT.P) H = P                                                      LNG1172
ASSIGN 3 TO KK                                                         LNG1173
1 CALL BETA(P,K1,T,V,NC)                                              LNG1174
K1 = H * K1                                                            LNG1175
VT = V + ((0.5E+00) *K1)                                              LNG1176
PT = P + ((0.5E+00)*H)                                                LNG1177
CALL BETA(PT,K2,T,VT,NC)                                              LNG1178
K2 = K2 * H                                                            LNG1179
VT = V + ((0.5E+00) *K2)                                              LNG1180
CALL BETA(PT,K3,T,VT,NC)                                              LNG1181
K3 = K3 * H                                                            LNG1182
PT = P + H                                                            LNG1183
VT = V + ( K3)                                                        LNG1184
CALL BETA(PT,K4,T,VT,NC)                                              LNG1185
K4 = K4 * H                                                            LNG1186
V = V + (((K1 + (2.E+00*K2) + (2.E+00*K3) + K4) / (6.E+00)) )      LNG1187
P = P - H                                                             LNG1188
IF(P.EQ.(0.0)) GO TO 4                                                LNG1189
IF(P.LT.(0.0)) GO TO 2                                                LNG1190
GO TO 1                                                                LNG1191
2 GO TO KK , (3,4)                                                    LNG1192
3 P = P + H                                                            LNG1193
H = P                                                                  LNG1194
ASSIGN 4 TO KK                                                         LNG1195
GO TO 1                                                                LNG1196
4 A = P                                                                LNG1197

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B = V	LNG1198
C = T	LNG1199
RETURN	LNG1200
END	LNG1201
SUBROUTINE TCM(TMC)	LNG1202
TYPE REAL NUM	LNG1203
TYPE INTEGER CNT1,CNT2,SWITCH	LNG1204
TYPE REAL MWM,INCR,KIJ,MW,MOL,LAM,LAMB	LNG1205
DIMENSION VCI(6),TH(12)	LNG1206
COMMON /RUN/ID(12),X(12),NAM(2,12),C12(12,12),R(12),Z(12),S(12),VS	LNG1207
1(12),TS(12),US(12),SG(12),EP(12),LAM(12),NC,MW(12),AIJ(12,12),PST(	LNG1208
212),TCT(12)	LNG1209
COMMON /DAT/TBI(12),VBI(12),TBIM(12),VBIM(12),C(12),SGIM(12),SIM(1	LNG1210
12),EPIM(12),PIM(12),SIN(12),VSIM(12),USIM(12),TSIM(12),DVBIM(12),P	LNG1211
2D(12),PV(12),RHO(12),VBIMP(12),TTM(12),MWM,CNT1,CNT2,TOLD(12),INCR	LNG1212
3(12),SWITCH,JPC,JPCS,JMIX,DENS,VEX,HEX,GEX,TMP,VMP,TP	LNG1213
COMMON/PAR/KIJ( 8, 8),AJI( 8, 8)	LNG1214
COMMON/UNITS/ITC,IPC,T,TMAX,DT,P,PMAX,DP,PS	LNG1215
DATA VCI/ 1.44,1.59,2.27,3.18,4.03,4.21/	LNG1216
DO 1 I = 1,NC	LNG1217
J = ID(I)	LNG1218
V = VCI(J)**(2./3.)	LNG1219
TH(I) = X(I) * V	LNG1220
1 CONTINUE	LNG1221
TTZ = 0.0	LNG1222
DO 2 I = 1,NC	LNG1223
TTZ = TTZ + TH(I)	LNG1224
2 CONTINUE	LNG1225
DO 3 I = 1,NC	LNG1226
TH(I) = TH(I) / TTZ	LNG1227
3 CONTINUE	LNG1228
SUM1 = 0.0	LNG1229
SUM2 = 0.0	LNG1230
K = NC - 1	LNG1231
DO 5 I = 1,K	LNG1232
L = I + 1	LNG1233
DO 4 J = L,NC	LNG1234
TTZ = (TCT(I)-TCT(J))/(TCT(I)+TCT(J))	LNG1235
TTZ= ABS(TTZ)	LNG1236
T12 = ((((((TTZ*(-3.038)))+(5.443))*TTZ)+(-1.343))*TTZ)	LNG1237
1 + (0.287))*TTZ) - (.0076)	LNG1238
T12 = T12*(TCT(I) + TCT(J)) * (0.9)	LNG1239
SUM2 = SUM2 + ((2.)*TH(I)*TH(J)*T12)	LNG1240
4 CONTINUE	LNG1241
SUM1 = SUM1 + (TH(I)*TCT(I)*(1.8))	LNG1242
5 CONTINUE	LNG1243
TMC = SUM1 + SUM2 + (TH(NC)*TCT(NC)*(1.8))	LNG1244
V = 0.0	LNG1245
DO 6 I = 1,NC	LNG1246
	LNG1247
J = ID(I)	LNG1248
TH(I) = X(I) * VCI(J)	LNG1249
V = V + TH(I)	LNG1250
6 CONTINUE	LNG1251
DO 7 I = 1,NC	LNG1252
TH(I) = TH(I) / V	LNG1253
7 CONTINUE	LNG1254
TMM = 0.0	LNG1255
DO 9 I = 1,NC	LNG1256
DO 8 J = 1,NC	LNG1257
M = ID(I)	LNG1258
N = ID(J)	LNG1259
NUM = (VCI(M)**(1./3.)) * (VCI(N)**(1./3.))	LNG1260



NUM = SQRT(NUM)	LNG1261
DEN = (0.5)*((VCI(M)**(1./3.))+(VCI(N)**(1./3.)))	LNG1262
NUM = NUM / DEN	LNG1263
NUM = NUM**(3.)	LNG1264
AKIJ = (1.0) - NUM	LNG1265
TCIJ = (1.0 - AKIJ) * SQRT(TCT(I)*TCT(J)*1.8*1.8)	LNG1266
TMM = TMM + TH(I)*TH(J)*TCIJ	LNG1267
8 CONTINUE	LNG1268
9 CONTINUE	LNG1269
TCMP = TMM + (10.0)	LNG1270
ICT = 0	LNG1271
TPO = TMM + (10.0)	LNG1272
10 TR = (T*1.8)/TCMP	LNG1273
NUM = (2901.01) - ((5738.92)*TR) + ((2849.85)*TR*TR)	LNG1274
1 + ((1.74127)/(1.01 - TR))	LNG1275
NUM = NUM * (TR - (1.0))	LNG1276
IF(NUM.LT.(-180.)) DD = 0	LNG1277
IF(NUM.LT.(-180.)) GO TO 11	LNG1278
DD = EXP(NUM)	LNG1279
C THE STATEMENT ABOVE MAY RESULT IN AN UNDERFLOW SENSE LIGHT	LNG1280
C ON SOME OPERATING SYSTEMS WHEN THE NUMBER NUM IS A LARGE	LNG1281
C NEGATIVE NUMBER. THE LARGE NEGATIVE VALUE IS PROPER AND	LNG1282
C THE CORRECT ANSWER FOR DD IS ZERO.	LNG1283
11 TCMP = TMM + ((TMC-TMM)*DD)	LNG1284
TEST = (1.0) - (TCMP/TPO)	LNG1285
TEST = ABS(TEST)	LNG1286
IF(TEST.LT.(.0001)) GO TO 12	LNG1287
ICT = ICT + 1	LNG1288
	LNG1289
IF(ICT.GT.250) GO TO 12	LNG1290
TPO = TCMP	LNG1291
GO TO 10	LNG1292
12 TMC = TCMP/(1.8)	LNG1293
RETURN	LNG1294
END	LNG1295
SUBROUTINE BLOCK	LNG1296
TYPE REAL KIJ,AJI	LNG1297
DIMENSION KIJ(8,8),AJI(8,8)	LNG1298
COMMON /PAR/ KIJ,AJI	LNG1299
DATA KIJ/	LNG1300
*0.000000,.293E-07,.911E-06,.0030498,0.001000,0.001000,0.001000,	LNG1301
*0.001000,1.088608,0.000000,.188E-10,.463E-06,.462E-08,.120E-03,	LNG1302
*0.001000,0.001000,1.098880,1.078910,0.000000,.517E-06,.535E-06,	LNG1303
*.188E-10,0.001000,0.001000,.8578232,1.146026,1.009559,0.000000,	LNG1304
*.546E-06,.136E-05,0.001000,0.001000,0.995000,1.287762,.9868486,	LNG1305
*.9690485,0.000000,.541E-06,0.001000,0.001000,1.178993,1.266327,	LNG1306
*1.016231,.9976706,.9435714,0.000000,0.001000,0.001000,0.950000,	LNG1307
*1.000000,1.000000,1.000000,1.000000,1.000000,0.000000,0.001000,	LNG1308
*0.995000,1.000000,1.000000,1.000000,1.000000,1.000000,1.000000,	LNG1309
*0.000000/	LNG1310
DATA AJI/	LNG1311
*0.000000,.161E-04,.257E-06,.0003462,0.001000,0.001000,0.001000,	LNG1312
*0.001000,1.014967,0.000000,.188E-10,.197E-06,.184E-05,.335E-06,	LNG1313
*0.001000,0.001000,1.008061,1.004105,0.000000,.142E-06,.103E-04,	LNG1314
*.188E-10,0.001000,0.001000,.9863463,1.011399,1.001862,0.000000,	LNG1315
*.105E-04,.119E-05,0.001000,0.001000,0.995000,1.024894,1.003289,	LNG1316
*.9987221,0.000000,.119E-05,0.001000,0.001000,.9566020,1.025269,	LNG1317
*1.005949,1.001225,.9968984,0.000000,0.001000,0.001000,0.995000,	LNG1318
*1.000000,1.000000,1.000000,1.000000,1.000000,0.000000,0.001000,	LNG1319
*0.995000,1.000000,1.000000,1.000000,1.000000,1.000000,1.000000,	LNG1320
*0.000000/	LNG1321
END	LNG1322



U.S. DEPT. OF COMM. <b>BIBLIOGRAPHIC DATA SHEET</b> (See instructions)		1. PUBLICATION OR REPORT NO. NBS TN-1030	2. Performing Organ. Report No.	3. Publication Date December 1980
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5. AUTHOR(S) Robert D. McCarty				
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10. SUPPLEMENTARY NOTES  <input type="checkbox"/> Document describes a computer program: SF-185, FIPS Software Summary, is attached.				
11. ABSTRACT (A 200-word or less factual summary of most significant information. If document includes a significant bibliography or literature survey, mention it here)  Four mathematical models of the equation of state for LNG like mixtures are presented. The four models include an extended corresponding states model, a cell model, a hard sphere model and a revised Klosek and McKinley model. Each of the models has been optimized to the same experimental data set which included data for pure nitrogen, methane, ethane, propane, iso and normal butane, iso and normal pentane and mixtures thereof. For LNG like mixtures (mixtures of the orthobaric liquid state at temperatures of 120 K or less and containing at least 60% methane, less than 4% nitrogen, less than 4% each of iso and normal butane and less than 2% total of iso and normal pentane), all of the models are estimated to predict densities to within 0.1% of the true value. The revised Klosek and McKinley model is valid only for mixtures within the range of temperature and composition specified above while the other three models are valid for a broader range of pressure, temperature and composition. The experimental PVTx data set used in the optimization together with comparisons are given and listings of computer programs for each of the models are included.				
12. KEY WORDS (Six to twelve entries; alphabetical order; capitalize only proper names; and separate key words by semicolons) Cell model; comparisons; computer programs; corresponding states; equation of state; hard sphere; LNG; mixtures; PVTx data; revised Kosek and McKinley.				
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